

- 1 In a test on some liquid fuels, decane is completely burnt in air.



What mass of oxygen would combine with 1 mol of decane?

- A** 15.5 g
- B** 248 g
- C** 496 g
- D** 672 g

Your answer

[1]

- 2 What volume of 0.250 mol dm⁻³ sodium hydroxide solution should be diluted to 1000 cm³ to make a 0.0100 mol dm⁻³ solution?

- A** 40 cm³
- B** 50 cm³
- C** 80 cm³
- D** 160 cm³

Your answer

[1]

- 3 A student carries out a titration. Sodium hydroxide solution is transferred to a conical flask using a pipette. Methyl orange indicator is added to the flask. Hydrochloric acid is added from a burette until the indicator changes colour.

Which of the following would lead to the titre being larger than it should be?

- A** Rinsing the conical flask with water before adding the sodium hydroxide solution.
- B** Rinsing the burette with water before filling it with hydrochloric acid.
- C** Rinsing the pipette with water before filling it with sodium hydroxide solution.
- D** Adding extra drops of indicator.

Your answer

[1]

A2 Chemistry Ocr Rates Practical Exam Quantitative

Frank Harriss



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