



IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match the following terms with their definitions.

Column A

- 1. ionic bond
- 2. metallic bond
- 3. covalent bond
- 4. ionic lattice
- 5. ionic radius
- 6. ionic radius
- 7. ionic radius
- 8. ionic radius
- 9. ionic radius
- 10. ionic radius
- 11. ionic radius
- 12. ionic radius
- 13. ionic radius
- 14. ionic radius
- 15. ionic radius
- 16. ionic radius
- 17. ionic radius
- 18. ionic radius
- 19. ionic radius
- 20. ionic radius

Column B

- a. the force of attraction between ions of opposite charge.
- b. the force of attraction between the positive nucleus and the negative electron.
- c. the force of attraction between the positive nucleus and the negative electron.
- d. the force of attraction between the positive nucleus and the negative electron.
- e. the force of attraction between the positive nucleus and the negative electron.
- f. the force of attraction between the positive nucleus and the negative electron.
- g. the force of attraction between the positive nucleus and the negative electron.
- h. the force of attraction between the positive nucleus and the negative electron.
- i. the force of attraction between the positive nucleus and the negative electron.
- j. the force of attraction between the positive nucleus and the negative electron.
- k. the force of attraction between the positive nucleus and the negative electron.
- l. the force of attraction between the positive nucleus and the negative electron.
- m. the force of attraction between the positive nucleus and the negative electron.
- n. the force of attraction between the positive nucleus and the negative electron.
- o. the force of attraction between the positive nucleus and the negative electron.
- p. the force of attraction between the positive nucleus and the negative electron.
- q. the force of attraction between the positive nucleus and the negative electron.
- r. the force of attraction between the positive nucleus and the negative electron.
- s. the force of attraction between the positive nucleus and the negative electron.
- t. the force of attraction between the positive nucleus and the negative electron.
- u. the force of attraction between the positive nucleus and the negative electron.
- v. the force of attraction between the positive nucleus and the negative electron.
- w. the force of attraction between the positive nucleus and the negative electron.
- x. the force of attraction between the positive nucleus and the negative electron.
- y. the force of attraction between the positive nucleus and the negative electron.
- z. the force of attraction between the positive nucleus and the negative electron.

B. Multiple Choice

Circle the letter of the best answer for each question.

- 1. Ionic bonding is best described as a(n) _____.
- a. covalent bond
- b. metallic bond
- c. ionic bond
- d. ionic lattice

ANSWER

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Kenneth W. Raymond

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