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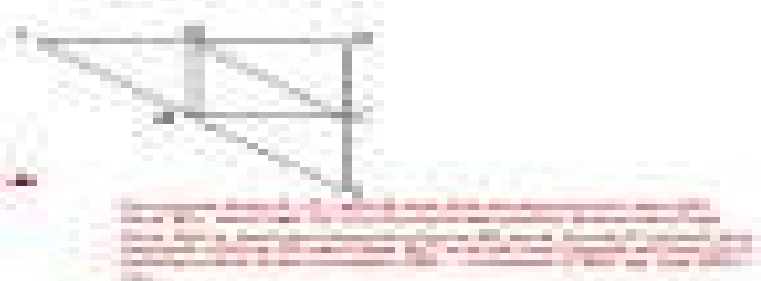
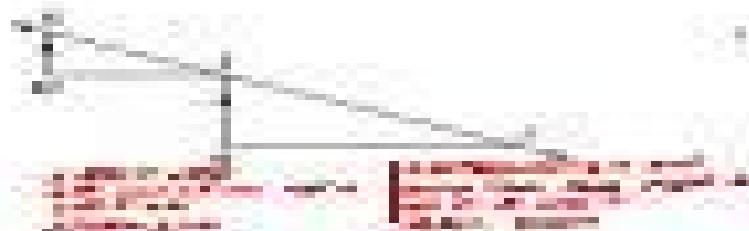


Figure 1. The effect of the concentration of the solution on the rate of the reaction. The rate of the reaction was measured at different concentrations of the solution. The rate of the reaction was found to be 0.0012 s⁻¹ at 0.1 M, 0.0024 s⁻¹ at 0.2 M, 0.0036 s⁻¹ at 0.3 M, 0.0048 s⁻¹ at 0.4 M, 0.0060 s⁻¹ at 0.5 M, 0.0072 s⁻¹ at 0.6 M, 0.0084 s⁻¹ at 0.7 M, 0.0096 s⁻¹ at 0.8 M, 0.0108 s⁻¹ at 0.9 M, and 0.0120 s⁻¹ at 1.0 M.

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