

Practice Exam 2

Ch 6 & 7 (Chaps. 1, 2, 3, 4, 5)

- 1.** Which of the following is a physical change?
 (a) rusting of metal (b) fruits ripen (c) baking a cake
 (d) ice melts (e) milk turns sour
- 2.** Which of the following is a homogeneous mixture?
 (a) water and sand (b) sand, alcohol, and sugar
 (c) oil and vinegar (d) milk of magnesium (e) air
- 3.** Which of the following is a correct?
 (a) sodium, S (b) mercury, Mg (c) iron, K, Sn
 (d) potassium, P (e) gold, Au
- 4.** What is the equivalent temperature for 105 Fahrenheit in Kelvin?
 (a) 41 (b) 314 (c) 378 (d) 457 (e) -32
- 5.** A sample of carbon tetrachloride has a mass of 39.75 g and a volume of 25.0 mL. What is its density (g/mL)?
 (a) 1.59 (b) 1.74 (c) 2.36 (d) 2.89 (e) 3.50 $d = \frac{m}{V} = \frac{39.75}{25.0 \text{ mL}}$
- 6.** How many significant figures are there in the measurement of 1.0056×10^7 ?
 (a) 1 (b) 2 (c) 3 (d) 4 (e) 5
- 7.** How many significant figures are appropriate to show in the result after carrying out the operation below?

$$(13.7 + 0.027) + 8.221 \rightarrow \text{X, } \div \text{ (last s.f.) } +, - \text{ (last decimal)}$$

 (a) 1 (b) 2 (c) 3 (d) 4 (e) 5
- 8.** How many protons, electrons, and neutrons are there in $^{120}\text{Ba}^{2+}$?
 (a) 56p, 56e, 130n (b) 56p, 54e, 130n (c) 56p, 58e, 74n
 (d) 56p, 54e, 74n (e) 54p, 56e, 74n
- 9.** Which of the following pairs would have similar chemical and physical properties?
 (a) Na, Mg (b) H, Li (c) He, Ne (d) Cr, Ca (e) Cl, O
- 10.** Which of the following is the molecular formula?
 (a) NaCl (b) CsF (c) HCl (d) AlBr₃ (e) NaCN
- 11.** Which of the following is the empirical formula?
 (a) C₂H₆ (b) CH₃COOH (c) P₂O₅ (d) NH₄NO₃ (e) Ag₂CO₃
- 12.** Which of the following is a correct match?
 (a) BaCl₂, barium dichloride (b) CS₂, carbon sulfide
 (c) CuF₂, copper (II) fluoride (d) HNO₃, nitrous acid
 (e) MgSO₄·7H₂O, magnesium sulfate heptahydrate
- 13.** Which of the following pairs is a correct match?
 (a) Ca₃(PO₄)₂, calcium (II) phosphate
 (b) (NH₄)₂SO₄, ammonium sulfate
 (d) Cu(NO₃)₂, copper dinitrate (c) Mg₂O₃, magnesium oxide
 (e) H₂CO₃, carbon acid

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Clifford Lane

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