

Chapter 5 The Periodic Table

Section 5.1 Organizing the Elements

(pages 126–129)

This section explains how elements organized elements into a periodic table. It also discusses the predictions he made about undiscovered elements and later the discovery of those elements supported his version of the table of the table.

Reading Strategy (page 126)

Identifying Main Ideas As you read, complete the table by identifying the main idea for each topic. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Topic	Main Idea
Mendeleev's proposal	Mendeleev arranged the elements in order of increasing mass so that elements with similar properties were in the same column.
Mendeleev's prediction	Mendeleev used the properties of existing elements to predict properties of undiscovered elements.
Evidence supporting Mendeleev's table	The close match between Mendeleev's predictions and the actual properties of new elements showed that his periodic table could be.

The Search for Order (page 126)

1. Is the following sentence true or false? The first elements to be identified were mainly gases. false
2. As the number of known elements grew, so did the need to organize them into groups based on their properties.
3. Circle the letter of each category that the French chemist Antoine Lavoisier used to classify elements.

☒ a. gases
c. liquids

☒ b. metals
☒ d. nonmetals

Mendeleev's Periodic Table (pages 127–128)

4. Is the following sentence true or false? Mendeleev needed to organize information about 63 elements. is so
5. Mendeleev's strategy for classifying elements was modeled on a(n) card game or solitaire.
6. Circle the letter of each type of information Mendeleev knew about each element.

☒ a. name

☐ b. number of protons

☒ c. relative mass

☒ d. properties

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