

Acids and Bases

Neutralization Reactions Lab



Reading, Lab, and Practice Bundle (Grades 6-9)

Acids and Bases

What is a Neutralization?

In neutralization reactions, an acid and base react to form a salt and water.

The type of chemical reaction important in neutralizing the pH of swimming pool water involves an acid reacting with a base to produce a balanced pH again.

What Are Acids and Bases?

Acids are substances that react with bases to form a low pH.

Examples: Lemon juice



Matching Vocabulary



Directions: Match each word on the left with the correct definition. Write the correct letter next to each number.

- (1) An substance with a pH greater than 7 that produces OH⁻ ions.
- (2) The product of an acid and base reaction. Often found in the form of NaCl.
- (3) A substance that changes color to show if a substance is acidic or basic.
- (4) A substance with a pH less than 7 that produces H⁺ ions.
- (5) A reaction between an acid and a base to form water and salt.

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28 Pages Printable & Easy to Use Includes Answer Key

Answer Key Neutralization Reactions Chapter 21 1

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