

Acids and Bases

Neutralization Reactions Lab



Reading, Lab, and Practice Bundle (Grades 6-9)

Acids and Bases

What is a Neutralization

A neutralization reaction is when an acid reacts with a base to form a salt and water.

This type of chemical reaction is important in everyday life - it's how our stomachs get rid of excess acid and pH is getting better.

What are Acids and Bases

Acids are substances that taste sour and have a low pH.

Examples: lemon juice, vinegar, etc.

Matching Vocabulary

Directions: Match each word on the left with the correct definition on the right. Write the correct letter next to each number.

- 1) A substance with a pH greater than 7 that produces OH^- ions. (A)
- 2) The product of an acid and base reaction, often found in the form of NaCl . (B)
- 3) A scale from 0 to 14 used to measure how acidic or basic a substance is. (C)
- 4) A chemical that changes color to show if a substance is acidic or basic. (D)
- 5) A substance with a pH less than 7 that produces H^+ ions. (E)
- 6) A reaction between an acid and a base to form water and salt. (F)

Matching Vocabulary

NAME: _____

DATE: _____

Directions: Match each word on the left with the correct definition or example on the right. Write the correct letter next to each number.

- 1) A substance with a pH greater than 7 that produces OH^- ions. (A) Acid
- 2) The product of an acid and base reaction, often found in the form of NaCl . (B) Base
- 3) A scale from 0 to 14 used to measure how acidic or basic a substance is. (C) pH scale
- 4) A chemical that changes color to show if a substance is acidic or basic. (D) Neutralization
- 5) A substance with a pH less than 7 that produces H^+ ions. (E) Salt
- 6) A reaction between an acid and a base to form water and salt. (F) Indicator

28 Pages Printable & Easy to Use Includes Answer Key

Answer Key Neutralization Reactions Chapter 21 1

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Staley, Michael S. Matta, Edward L.
Waterman, Prentice-Hall Staff**



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