

# ANSWER KEY - ORGANIZED BY REACTION TYPE

Practice Box	Practice Box	Practice Box
Part A	Part B	Part C: Double Replacement Reactions
<p>Instructions: Balance each of the following chemical equations. Write the balanced equations in the boxes below. Then, check your answers with the answer key at the end of this section.</p> <p><b>Part A:</b></p> <ol style="list-style-type: none"> <li><math>Ca + O_2 \rightarrow CaO</math></li> <li><math>Na_2O + H_2O \rightarrow NaOH</math></li> <li><math>Ca + H_2O \rightarrow Ca(OH)_2 + H_2</math></li> </ol> <p><b>Part B:</b></p> <ol style="list-style-type: none"> <li><math>Na + H_2O \rightarrow NaOH + H_2</math></li> </ol> <p><b>Part C: Double Replacement Reactions</b></p> <ol style="list-style-type: none"> <li><math>NaOH + HCl \rightarrow NaCl + H_2O</math></li> </ol> <p><b>ANSWER KEY:</b></p> <p>Part A: Combination Reactions</p> <ol style="list-style-type: none"> <li><math>Ca + O_2 \rightarrow 2CaO</math></li> <li><math>2Na + 2H_2O \rightarrow 2NaOH + H_2</math></li> </ol> <p>Part B: Single Replacement Reactions</p> <ol style="list-style-type: none"> <li><math>2Na + 2H_2O \rightarrow 2NaOH + H_2</math></li> </ol> <p>Part C: Double Replacement Reactions</p> <ol style="list-style-type: none"> <li><math>NaOH + HCl \rightarrow 2NaCl + H_2O</math></li> </ol>		

Practice Box	Practice Box	Practice Box
Part D: Double Replacement Reactions	Part E: Single Replacement Reactions	Part F: Double Replacement Reactions
<p>Instructions: Balance each of the following chemical equations. Write the balanced equations in the boxes below. Then, check your answers with the answer key at the end of this section.</p> <p><b>Part D:</b></p> <ol style="list-style-type: none"> <li><math>NaOH + HCl \rightarrow NaCl + H_2O</math></li> </ol> <p><b>Part E:</b></p> <ol style="list-style-type: none"> <li><math>Na + H_2O \rightarrow NaOH + H_2</math></li> </ol> <p><b>Part F: Double Replacement Reactions</b></p> <ol style="list-style-type: none"> <li><math>NaOH + HCl \rightarrow NaCl + H_2O</math></li> </ol> <p><b>ANSWER KEY:</b></p> <p>Part D: Double Replacement Reactions</p> <ol style="list-style-type: none"> <li><math>NaOH + HCl \rightarrow NaCl + H_2O</math></li> </ol> <p>Part E: Single Replacement Reactions</p> <ol style="list-style-type: none"> <li><math>Na + H_2O \rightarrow NaOH + H_2</math></li> </ol> <p>Part F: Double Replacement Reactions</p> <ol style="list-style-type: none"> <li><math>NaOH + HCl \rightarrow NaCl + H_2O</math></li> </ol>		

ANSWER KEY
<p><b>Part A: Combination Reactions</b></p> <ol style="list-style-type: none"> <li><math>Ca + O_2 \rightarrow 2CaO</math></li> <li><math>2Na + 2H_2O \rightarrow 2NaOH + H_2</math></li> </ol> <p><b>Part B: Single Replacement Reactions</b></p> <ol style="list-style-type: none"> <li><math>2Na + 2H_2O \rightarrow 2NaOH + H_2</math></li> </ol> <p><b>Part C: Double Replacement Reactions</b></p> <ol style="list-style-type: none"> <li><math>NaOH + HCl \rightarrow 2NaCl + H_2O</math></li> </ol> <p><b>Part D: Decomposition Reactions</b></p> <ol style="list-style-type: none"> <li><math>2Na_2O_2 + 2H_2O \rightarrow 2Na_2O + O_2</math></li> <li><math>Na_2O_2 + H_2O \rightarrow Na_2O + H_2O</math></li> <li><math>2Na_2O_2 + 2H_2O \rightarrow 2Na_2O + O_2</math></li> <li><math>Na_2O_2 + H_2O \rightarrow Na_2O + H_2O</math></li> <li><math>Na_2O_2 + H_2O \rightarrow Na_2O + H_2O</math></li> <li><math>Na_2O_2 + H_2O \rightarrow Na_2O + H_2O</math></li> </ol>

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