

ALTERNATIVE B

Maximum marks obtainable: 50 marks

Answer all the questions

Question 1

(a) Two concordant titres = 8 marks
 Averaging = 2 marks
 [10 marks]



(Accept $\text{Na}_2\text{CO}_3 + 2\text{HNO}_3 \rightarrow 2\text{NaNO}_3 + \text{CO}_2 + \text{H}_2\text{O}$)

(c) (i) Concentration of H in mol dm^{-3}

$$\text{Mass of HNO}_3 \text{ in } 1 \text{ dm}^3 = \frac{6.3 \times 1000}{500} \quad (1) \\ = 12.60 \text{ g} \quad (1) \text{ (Accept 12.6g)}$$

$$\text{Molar mass of HNO}_3 = 1 + 14 + (16 \times 3) = 63 \text{ g mol}^{-1} \quad (1) \text{ (no score for wrong unit)} \\ \text{Conc. HNO}_3 = \frac{12.6}{63} \quad (1) = 0.20 \text{ mol dm}^{-3} \quad (1) \text{ (Accept 0.2 mol dm}^{-3})$$

$$\frac{\text{C}_A \text{V}_A}{\text{C}_B \text{V}_B} = \frac{2}{1} \quad (1) \text{ (Molar ratio)}$$

$$\text{C}_B = \frac{\text{C}_A \text{V}_A}{2 \times \text{V}_B} \quad (1) \text{ (Making C}_B \text{ the subject)}$$

$$= \frac{0.2 \times \text{V}_A}{2 \times \text{V}_B} \quad (1) \text{ (Correct substitution)}$$

$$= \frac{S}{2} \text{ say} \quad (1) \text{ (Correct evaluation to 3 sig. fig.)} \\ \text{(Wrong unit, no score)}$$

(ii) Mass of $\text{Na}_2\text{CO}_3 \cdot \text{xH}_2\text{O}$ in $\text{dm}^3 = \frac{8.20 \times 1000}{250} \quad (1) = 32.8 \text{ g/dm}^3 \quad (1)$

$$\text{Molar mass of } \text{Na}_2\text{CO}_3 \cdot \text{xH}_2\text{O} = \frac{\text{Mass conc.}}{\text{Molar conc.}} = \frac{32.8}{S} \quad (1)$$

$$= Q \text{ say} \quad (1) \quad (\text{Accept to 2 decimal places})$$

Answer For Chemistry Practical Alternative B

William S. Furneaux

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Answers to questions set at the science and art examinations Science and art department,1887 UGC-NET Forensic Science Solved Previous year Question Paper Book With Solution [Year 2019 to 2024] As Per Updated Syllabus ,2025-05-01
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