

13. (4 pts) What is an ionic compound?
a. A compound that will dissolve in water
b. Two nonmetal atoms bonded together
c. Two metal atoms bonded together
d. A metal and a nonmetal atom bonded together
e. None of the above

14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
a. Ionic
b. Covalent
c. Both a and b
d. Neither a nor b

15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
a. Ionic
b. Covalent
c. Both a and b
d. Neither a nor b

16. (4 pts) How many total atoms are in this molecule: $\text{C}_6\text{H}_{12}\text{OH}$?
a. 3
b. 4
c. 19
d. 20
e. None of the above

17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is $\text{C}_x\text{H}_{10}\text{Cl}_3$, what must x be?
a. 1
b. 2
c. 3
d. 4
e. None of the above

18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water? $\text{C}_6\text{H}_5\text{ClF}$ $\text{C}_6\text{H}_5\text{CH}_3$ CS_2 BaSO_4
a. $\text{C}_6\text{H}_5\text{ClF}$
b. $\text{C}_6\text{H}_5\text{CH}_3$
c. CS_2
d. BaSO_4
e. CS_2 and BaSO_4

Apologia Chemistry Module 13 Test

H Kauffman

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Table of Contents Apologia Chemistry Module 13 Test

1. Understanding the eBook Apologia Chemistry Module 13 Test
 - The Rise of Digital Reading Apologia Chemistry Module 13 Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Apologia Chemistry Module 13 Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Apologia Chemistry Module 13 Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Apologia Chemistry Module 13 Test
 - Personalized Recommendations
 - Apologia Chemistry Module 13 Test User Reviews and Ratings
 - Apologia Chemistry Module 13 Test and Bestseller Lists
5. Accessing Apologia Chemistry Module 13 Test Free and Paid eBooks
 - Apologia Chemistry Module 13 Test Public Domain eBooks
 - Apologia Chemistry Module 13 Test eBook Subscription Services
 - Apologia Chemistry Module 13 Test Budget-Friendly Options

6. Navigating Apologia Chemistry Module 13 Test eBook Formats
 - ePUB, PDF, MOBI, and More
 - Apologia Chemistry Module 13 Test Compatibility with Devices
 - Apologia Chemistry Module 13 Test Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Apologia Chemistry Module 13 Test
 - Highlighting and Note-Taking Apologia Chemistry Module 13 Test
 - Interactive Elements Apologia Chemistry Module 13 Test
8. Staying Engaged with Apologia Chemistry Module 13 Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Apologia Chemistry Module 13 Test
9. Balancing eBooks and Physical Books Apologia Chemistry Module 13 Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Apologia Chemistry Module 13 Test
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Apologia Chemistry Module 13 Test
 - Setting Reading Goals Apologia Chemistry Module 13 Test
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Apologia Chemistry Module 13 Test
 - Fact-Checking eBook Content of Apologia Chemistry Module 13 Test
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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