

Neutralization Reaction

TABLE 21.1 Acid–Base Reactions and the Salts Formed

Acid		Base		Salt		Water
HCN Hydrogen cyanide	+	NaOH Sodium hydroxide	→	NaCN Sodium cyanide	+	H_2O
HNO_3 Nitric acid	+	KOH Potassium hydroxide	→	KNO_3 Potassium nitrate	+	H_2O
2HCl Hydrogen chloride	+	$\text{Ca}(\text{OH})_2$ Calcium hydroxide	→	CaCl_2 Calcium chloride	+	$2 \text{H}_2\text{O}$
HF Hydrogen fluoride	+	NaOH Sodium hydroxide	→	NaF Sodium fluoride	+	H_2O

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RS Peters

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