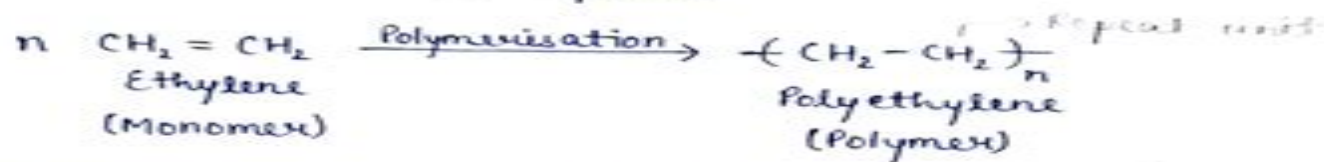


POLYMERS

Greek word $\begin{cases} \rightarrow \text{Poly} \rightarrow \text{many} \\ \rightarrow \text{Mer} \rightarrow \text{parts} \end{cases}$

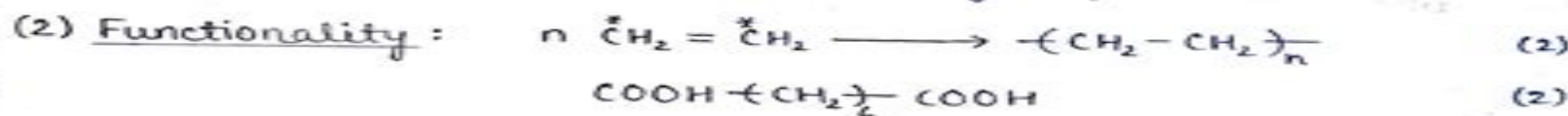


Polymers are macromolecules whose structure consists of large number of repeat units.

Characteristics of Polymers :-

- (1) Degree of Polymerisation (D_p): It is the number of monomers combined to form a polymer.

$$D_p = \frac{M \text{ (Molecular Weight of Polymer)}}{m \text{ (Molecular weight of monomer)}}$$



It is the number of bonding sites present in a monomer. For a substance to act as a monomer it should be atleast bifunctional.

- (i) If monomer is bifunctional, $F=2$ (linear chain polymer)
- (ii) If monomers are bifunctional and trifunctional, $F=2,3$ (Branched chain polymer)
- (iii) If all monomers are trifunctional, $F=3$ (Cross-linked chain polymer)

- (3) Molecular Weight: Polymers are poly-dispersed in nature so their molecular weight is calculated as an average.

- (i) Number - average molecular weight ($\bar{M}_n$)

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