

# CHEMISTRY MULTIPLE CHOICE QUESTIONS

## A. Atoms, molecules and stoichiometry

2002 -2014

### MEGA LECTURE

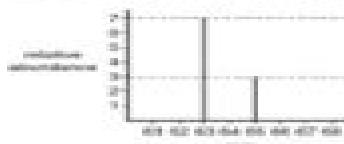
As a simplification, an adult human can be considered to have a body mass of 70,000 g of methachloro isopropyl bromide  $\text{CH}_2\text{ClCH}_2\text{CH}_3\text{Br}$ .

What mass of carbon dioxide does a person exhale each day if all the methachloro isopropyl bromide is oxidized and burned?

A. 0.007 kg      B. 0.0004 kg      C. 0.004 kg      D. 0.04 kg

(2002 MCQ (2))

The diagram shows the mass spectrum of a sample of naturally-occurring copper.



What is the relative atomic mass of this copper?

A. 63.0      B. 63.5      C. 63.8      D. 64.0

(2002 MCQ (2))

Which pairs of compounds have the same empirical formula?

1. ethane and ethene  
2. ethane and cyclohexane  
3. cyclohexane and cyclohexene

(2002 MCQ (2))

A mixture of  $10\text{cm}^3$  of methane and  $10\text{cm}^3$  of ethane was reacted with an excess of oxygen, after cooling to room temperature, the residual gas was passed through a porous palladium pallidometer.

Which compounds have the empirical formula  $\text{CH}_2\text{O}$ ?

A. methanol  
B. ethanoic acid  
C. methyl ethanoate

(2002 MCQ (2))

What is the number of molecules in  $500\text{ cm}^3$  of oxygen under room conditions?

A.  $1.25 \times 10^{24}$   
B.  $1.34 \times 10^{24}$   
C.  $3.0 \times 10^{24}$   
D.  $3.0 \times 10^{25}$

(2002 MCQ (2))

In the preparation of soft soap, glycerol is added.



In a laboratory experiment, it was found, on average, one of the isobutane is converted into the  $\text{CH}_2=\text{CHCH}_2\text{OH}$  and the  $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$  and  $\text{CH}_3\text{OH}$  molecules are converted into the  $\text{CH}_2=\text{CHCH}_2\text{OH}$  molecule.

How many moles of the glycerol are required to convert one mole of isobutane to isobutanol into the soft soap?

A. 0      B. 1      C. 10      D. 100

(2002 MCQ (2))

One of the Data Block is relevant to this question.

Stoichiometric mass: stoichiometric amounts of reactants needed to form 1 mol of product

# 2012 Hsc Chemistry Multiple Choice Answers

**Roger Slade, Maureen Slade**

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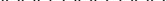
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