

## Chemistry 2<sup>nd</sup> Semester Final Exam Review

### Chemical Bonds

- Give a physical description of how the atoms and molecules are arranged in solids, liquids, and gases.  
*A: In a liquid, the forces between the molecules are weak, so the molecules are free to move around. In a solid the forces between the molecules (intermolecular forces) are stronger so they are held in place.*
- Use the information from the previous question to describe the shape and volume for each phase (state) of matter.  
*A: Solids have a repeating geometric pattern (lattice structure) in which the particles vibrate in place. They have a definite shape and a definite volume.  
Particles within a liquid slide past each and are close, but are not held in place. They have a definite volume, but no definite shape.  
Gases particles move chaotically bouncing off one another. They have neither a definite shape nor a definite volume.*
- List the three phases of matter in order of increasing intermolecular attractions.  
*A: Gas, liquid, solid.*
- Why do the atoms and molecules in liquids move in a random pattern relative to one another instead of being in a solid form?  
*A: Their kinetic energy (temperature) is high enough to overcome the attractive forces that might otherwise hold them in a repeating pattern. The attractive forces aren't strong enough to hold them together at that temperature.*

### Conservation of Matter and Stoichiometry

- The atomic mass unit and the mole are defined using what isotope?  
*A: Carbon-12*
- Balance the following equations:
  - $\text{Al}_2\text{O}_3 + \text{Cl}_2 + \text{C} \rightarrow \text{AlCl}_3 + \text{CO}$   
*A:  $\text{Al}_2\text{O}_3 + 3\text{Cl}_2 + 3\text{C} \rightarrow 2\text{AlCl}_3 + 3\text{CO}$*
  - $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   
*A:  $2\text{C}_2\text{H}_6 + 7\text{O}_2 \rightarrow 8\text{ CO}_2 + 6\text{H}_2\text{O}$*
- What isotope is used to define one mole?  
*A: Carbon-12*
- How many particles (atoms or molecules) are in one mole?  
*A:  $6.02 \times 10^{23}$*
- Find the molar mass for the following chemical formulas:
  - $\text{ZnF}_2$   
*A: 103.377 g/mol*
  - $\text{Al}_2(\text{SO}_4)_3$   
*A: 342.151 g/mol*
- How many moles are in 4.37 kg of NaOH?  
*A: 109.3 mol*
- How many atoms are in 2075 g of He?  
*A:  $3.12 \times 10^{26}$  atoms*
- What volume will 92.31 g of CO<sub>2</sub> occupy at STP?  
*A: 46.985 L*
- How many grams of sodium hydroxide are needed to react with 6.23 g of barium bromide?  
$$\text{NaOH} + \text{BaBr}_2 \rightarrow \text{NaBr} + \text{Ba(OH)}_2$$
  
*A: 1.68 g NaOH*
- In the reaction,  $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ , if 100.0 g of magnesium reacts with 50.0 g of oxygen, what mass of product is produced?  
*A: 126 g MgO*

### Gases and Their Properties

- What is the kinetic molecular theory (KMT)?  
*A: The Kinetic Molecular Theory is used to explain the behavior of gases and is based upon the following postulates:
  - Gas are composed of a smoky particles that behave like hard spherical objects in a state of constant, random motion.
  - These particles move in a straight line until they collide with another particle or the walls of the container.
  - These particles are much smaller than the distance between particles, therefore the volume of a gas is mostly empty space and the volume of the gas molecule themselves is negligible.
  - There is no force of attraction between gas particles or between the particles and the walls of the container.
  - Collisions between gas particles or collisions with the walls of the container are elastic. That is, none of the energy of the gas particle is lost in a collision.*

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**JG Myers**

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*The Effectiveness of a Computer-assisted-instruction Package in Supplementing Teaching of Selected Concepts in High School Chemistry* Camille Linda Wainwright,1984    *The American Secondary School* Leslie Owen Taylor,1960  
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*Cornell University Courses of Study* Cornell University,2003    **The School Review** ,1914    *Source Book for Chemistry Teachers* William Thomas Lippincott,1981    *Summer School Number* Southwestern at Memphis,1933    **Lucky Break** Howard Junker,1999 In **Lucky Break** Howard Junker introduces you to nineteen contemporary authors who describe their struggles and good fortune in transforming themselves into writers For some the idea of being a writer seemed to run in the family For others writing was a way of breaking with their families for still others writing was the only way to find their own voice to preserve what is uniquely theirs Collected from the pages of ZYZZYVA the West Coast literary journal these are very personal essays in which nevertheless other writers will discover something of their own stories and journeys Among the authors featured in **Lucky Break** are Sheila Ballantyne Opal Palmer Adisa Philip Levine Jewelle Gomez Justin Chin David Rains Wallace and Sallie Tisdale Becoming a writer is a mysterious momentous process The writers in this collection suggest that beyond talent and hard work and a lucky break there is more hard work the task of being a writer They offer accounts not of the craft of writing but of the many strands that must be woven together to create an identity as a writer    *Current Index to Journals in Education* ,1974    *Catalog Issue* University of Colorado,1974    *Zyzyva* ,1999    **Resources in Education** ,1978    *Paperbound Books in Print* ,1966    *American Journal of Physics* ,1994    **Science Books & Films** ,1980    *Computer Aided Training in Science and Technology* E. Oñate,1990    **Current Index to Journals in Education, Semi-Annual Cumulation, July-December, 1976** ,1976-10

The Top Books of the Year Nd Semester Chemistry Exam Review Fot Plato The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous captivating novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the fascinating narratives that have charmed audiences this year. The Must-Read : Colleen Hoovers "It Ends with Us" This poignant tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Nd Semester Chemistry Exam Review Fot Plato : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and thrilling novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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