

14.3

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Identify compounds based on the pH of their aqueous solutions and the pH of an aqueous solution based on the chemical formula and concentration of the compound and the chemical formula of the compound.
2. Calculate the pH of a solution (25°C).

Vocabulary

- | | | |
|-------------------|---|---------------------|
| 1. autoionization | 2. hydronium ion (H_3O^+) | 3. acid solution |
| 4. autoionization | 5. acid rain | 6. neutral solution |
| | 7. acid solution | 8. pH |

Key Equation

- | | |
|-------------------------------------|---|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}_3\text{O}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}_3\text{O}^+]$ |

Part A Completion

Write the requested number or chemical formula(s) of the compound(s) that are indicated in the space provided. Answers should include correct state symbols (aqueous or solid).

1. What substance(s) is/are _____, contains hydrogen ion(s)?
2. Write the chemical formula(s) of the compound(s) of those which are indicated in the space provided in _____.
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