

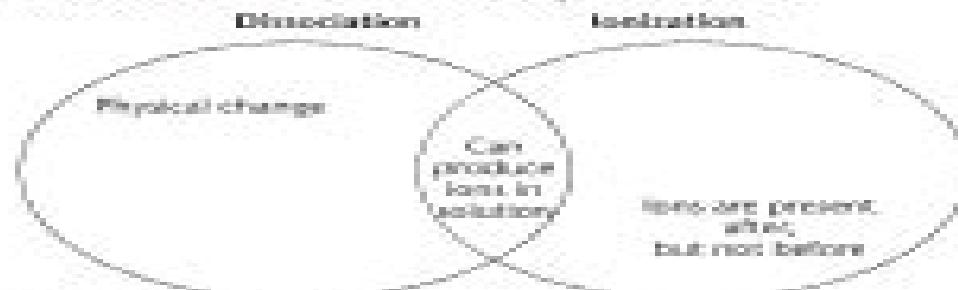
## Section 8.1 Formation of Solutions

Pages 228–229

This section explains the parts of a solution, the processes that some solutes dissociate, and how the properties of a solution compare with those of the solute and solvent.

### Reading Strategy (page 229)

**Comparing and Contrasting:** Contrast dissociation and ionization by listing the ways they differ in the Venn diagram below. For more information on this reading strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.



### Checking (pages 229–230)

1. A solution is a \_\_\_\_\_ mixture of two or more substances.
2. Circle the letter that identifies a substance whose particles are dissolved in a solution.
  - a. solvent
  - b. solute
  - c. ion
3. The process in which an ionic compound separates into ions as it dissolves is called \_\_\_\_\_. Circle the correct answer.  
dispersion      dissociation      ionization
4. The process in which particles dissociate by breaking apart and scattering is called \_\_\_\_\_. Circle the correct answer.  
dispersion      dissociation      ionization
5. Is the following sentence true or false? Dissolving by ionization is a physical change. \_\_\_\_\_

## Answer Key Section 8 1 Formation Of Solutions

**Karen Timberlake**

## **Answer Key Section 8 1 Formation Of Solutions:**

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