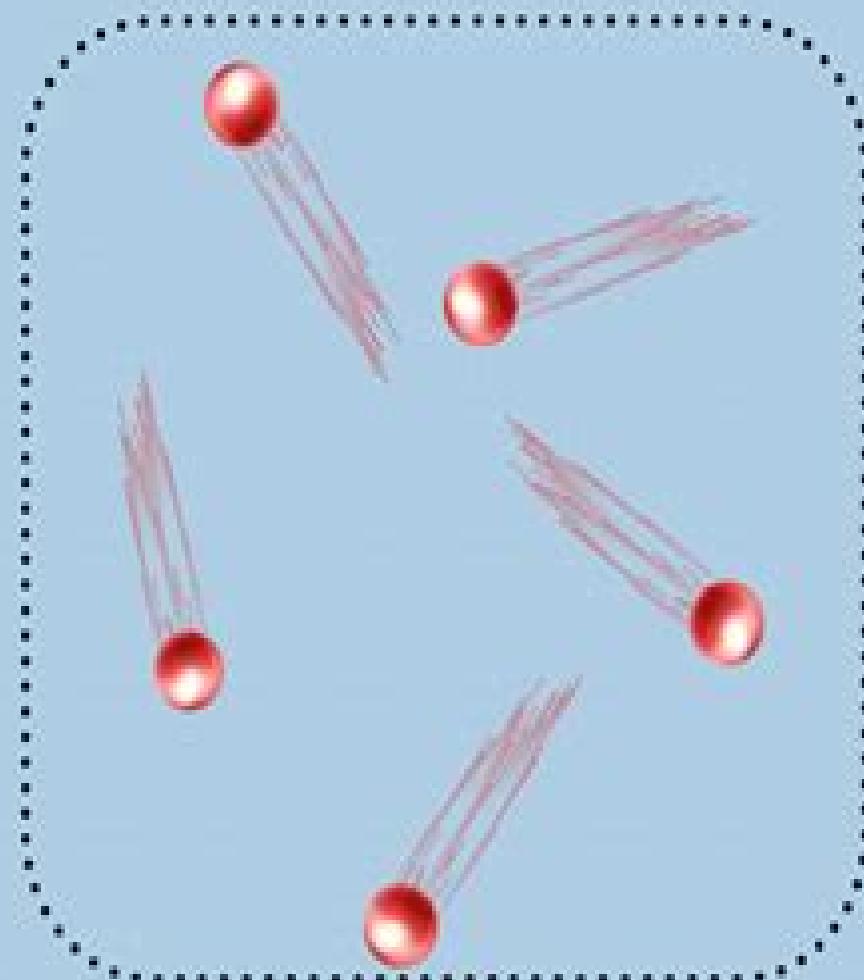


Kinetic Molecular Theory of Gases

- Many gas particles, so statistics applies
- Particles are tiny and spread far apart
- Gas particles are in constant random motion
- No attraction or repulsion between particles
- Collisions are elastic



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Chemistry

Rachel Sandford

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Zumdahl,2003-07 Contains chapter discussions key term definitions and practice chapter tests Study Guide,
Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry Iris Stovall,Donald J. DeCoste,2000
Investigations in Natural Science: pt.1. Chemistry. Teacher's guide ,1985 CHEMISTRY SILBERBERG,2003
Publications of the Department of Chemistry Ohio State University. Dept. of Chemistry,1937 Chemistry and Industry
,1962 **Research in Education** ,1970-05 Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998
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