

BACONG SILANG NATIONAL HIGH SCHOOL
1ST QUARTER EXAMINATION
PHYSICAL SCIENCE

NAME: _____
GRADE: _____

DATE: _____
SCORE: _____

Directions: Write the letter of the correct answer on the space provided.

1. It is also known as nuclear fusion and the formation of new nuclei actions in the early stages of development of the universe.
a. Nucleosynthesis b. S-Process c. R-Process d. Singularity

2-4. There are two categories of star, the average star and massive star. The life cycle of these two stars are common from 1st to 3rd stages namely nebula, protostar, and main sequence star. What will be the next stages in the life of an average star until it reaches its death?
a. Red super giant \rightarrow Supernova \rightarrow Black hole \rightarrow neutron star 0
b. Red giant \rightarrow white dwarf \rightarrow black dwarf 3
c. Red giant \rightarrow white dwarf \rightarrow black hole 2
d. Red supergiant \rightarrow white dwarf \rightarrow black dwarf 1

5. He synthesized element with atomic number 43 using a linear particle accelerator.
a. Ernest Rutherford b. Ernest Lawrence c. Dmitri Mendeleev d. John Dalton

6. The following are light elements EXCEPT:
a. helium b. hydrogen c. lithium d. iron

7. Which of the following molecule is nonpolar?
a. NaCl b. HCl c. CO_2 d. NH_3

8. What type of chemical bond holds the atoms of water molecule together?
a. Hydrogen bond b. ionic bond c. Polar covalent bond d. Nonpolar covalent bond

9. Which of the following IMFAs is considered as the weakest?
a. H-bonding b. Ion-dipole c. Dipole-dipole d. London forces

10. Which of the following statements is TRUE about nonpolar molecules?
a. high boiling point b. high melting point c. low surface tension d. low vapor pressure

11. Forces that are present between an ion and a polar molecule
a. Ion-dipole b. dipole-dipole c. dispersion forces d. hydrogen bonding

12. H-bonding forms when the substances involved are polar and there are _____.
a. H-atoms of a molecule interact with to O, N, F of other molecules.
b. C-atoms attached to O, N, F.
c. central atoms with O, N, F as attached atoms.
d. unshared pair of electrons in the central atoms.

13. The amount of energy required to stretch or increase the surface of a liquid by a unit area
a. specific heat b. surface tension c. vapor pressure d. heat of vaporization

14. Liquids can form spherical elastic film to minimize surface area. What intermolecular force is responsible for the formation of this film in water?
a. H-bonding b. ion-induced dipole c. dipole-induced dipole d. London dispersion force

15. Which group of biomolecules carries and passes on the hereditary information of the organism?
a. carbohydrates b. lipids c. nucleic acids d. proteins

16. What are the elements that make up carbohydrates?
a. C and H b. C, H and N c. C, H and O d. C, H, O and N

17-19. Is carbohydrates one of the most important biological macromolecules needed by a human body?
a. Yes, because it is the primary source of energy. 3
b. Yes, because it can be a storage of energy for a long period of time. 1
c. Yes, because it can be found in rice, cereals, fruits and pastas. 2
d. Yes, because it carries genetic information. 0

20. The measure of how fast is the change in the concentration of the reactants or products in a chemical reaction.
a. Activation Energy b. Activation Rate c. Chemical Reaction d. Concentration

21. A chemical change that occurs when two or more substances combine to form a new substance.
a. Activation Energy b. Chemical Kinetics c. Chemical Reaction d. Product

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Arthur James Wells

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Table of Contents Answer Key For Odyssey Physical Science

1. Understanding the eBook Answer Key For Odyssey Physical Science
 - The Rise of Digital Reading Answer Key For Odyssey Physical Science
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Odyssey Physical Science
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Odyssey Physical Science
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Odyssey Physical Science
 - Personalized Recommendations
 - Answer Key For Odyssey Physical Science User Reviews and Ratings
 - Answer Key For Odyssey Physical Science and Bestseller Lists
5. Accessing Answer Key For Odyssey Physical Science Free and Paid eBooks

- Answer Key For Odyssey Physical Science Public Domain eBooks
- Answer Key For Odyssey Physical Science eBook Subscription Services
- Answer Key For Odyssey Physical Science Budget-Friendly Options

6. Navigating Answer Key For Odyssey Physical Science eBook Formats

- ePUB, PDF, MOBI, and More
- Answer Key For Odyssey Physical Science Compatibility with Devices
- Answer Key For Odyssey Physical Science Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Answer Key For Odyssey Physical Science
- Highlighting and Note-Taking Answer Key For Odyssey Physical Science
- Interactive Elements Answer Key For Odyssey Physical Science

8. Staying Engaged with Answer Key For Odyssey Physical Science

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Answer Key For Odyssey Physical Science

9. Balancing eBooks and Physical Books Answer Key For Odyssey Physical Science

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Answer Key For Odyssey Physical Science

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Answer Key For Odyssey Physical Science

- Setting Reading Goals Answer Key For Odyssey Physical Science
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Answer Key For Odyssey Physical Science

- Fact-Checking eBook Content of Answer Key For Odyssey Physical Science
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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