

Name Solutions

Date _____ Period _____

Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



| Atoms | Valence electrons | Electron transfer from/to each atom | Ions formed in the product |
|-------|-------------------|-------------------------------------|----------------------------|
| Li | 1 ve | transfers $1e^-$ to F | Li^+ |
| F | 7 ve | accepts $1e^-$ from Li | F^- |

| Reactions | Atoms | Valence electrons | Electron transfer from/to each atom | Ions formed in the product |
|---|-------|-------------------|-------------------------------------|----------------------------|
| 1) $Li + Cl \Rightarrow LiCl$ $Li + Cl \rightarrow [Li]^+ [Cl]^-$ | Li | 1 ve | donate $1e^-$ | Li^+ |
| | Cl | 1 ve | accepts $1e^-$ | Cl^- |
| 2) $Ca + O \Rightarrow CaO$ $Ca + O \rightarrow [Ca]^{+2} [O]^{2-}$ | Ca | 2 ve | donates $2e^-$ | Ca^{+2} |
| | O | 6 ve | accepts $2e^-$ | O^{2-} |
| 3) $Be + F \Rightarrow BeF_2$ $Be + 2F \rightarrow [Be]^{+2} [F]_2^{2-}$ | Be | 2 ve | donates $2e^-$ to each F | Be^{+2} |
| | F | 7 ve | each F accepts $1e^-$ each | F^- |
| 4) $Mg + S \Rightarrow MgS$ $Mg + S \rightarrow [Mg]^{+2} [S]^{2-}$ | Mg | 2 ve | donates $2e^-$ to S | Mg^{+2} |
| | S | 6 ve | accepts $2e^-$ | S^{2-} |
| 5) $K + F \Rightarrow KF$ $K + F \rightarrow [K]^+ [F]^-$ | K | 1 ve | donates $1e^-$ | K^+ |
| | F | 7 ve | accepts $1e^-$ | F^- |

Answer Key For Ionic Bonding Worksheet

Jiyuan Zhang

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Answer Key For Ionic Bonding Worksheet:

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