

Unit 3 Worksheet 3 – Quantitative Energy Problems

Energy constants (H₂O)

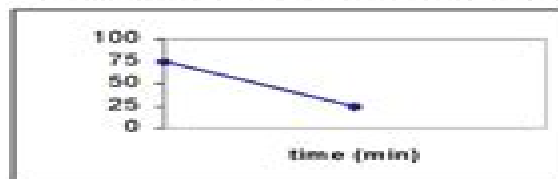
334 J/g	Heat of fusion (melting or freezing) H _f
2260 J/g	Heat of vaporization (evaporating or condensing) H _v
2.1 J/g°C	Heat capacity (c) of solid water
4.18 J/g°C	Heat capacity (c) of liquid water

$$Q = m \cdot c \cdot \Delta T$$

$$Q = m \cdot H_v \text{ or } m \cdot H_f$$

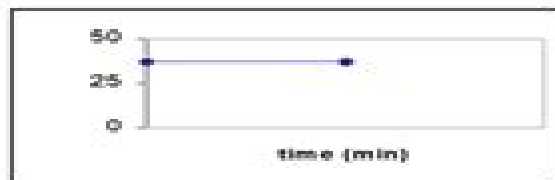
For each of the problems sketch a warming or cooling curve to help you decide which equation(s) to use to solve the problem. Keep a reasonable number of sig figs in your answers.

1. A cup of coffee (140 g) cools from 75°C down to comfortable room temperature 20.°C. How much energy does it release to the surroundings?



$$Q = 140g \cdot 4.18 \frac{J}{g^{\circ}C} \cdot 55^{\circ}C = 32,000 J \text{ or } 32 kJ$$

2. Suppose during volleyball practice, you lost 2.0 lbs of water due to sweating. If all of this water evaporated, how much energy did the water absorb from your body? Express your answer in kJ. 2.2 lbs = 1.0 kg



$$2.0 lbs \cdot \frac{1kg}{2.2 lbs} \cdot \frac{1000g}{1kg} = 910g$$

$$Q = 910g \cdot 2260 \frac{J}{g} = 2,100,000 J \text{ or } 2100 kJ$$

3. Suppose that during the Icy Hot lab that 65 kJ of energy were transferred to 450 g of water at 20°C. What would have been the final temperature of the water?

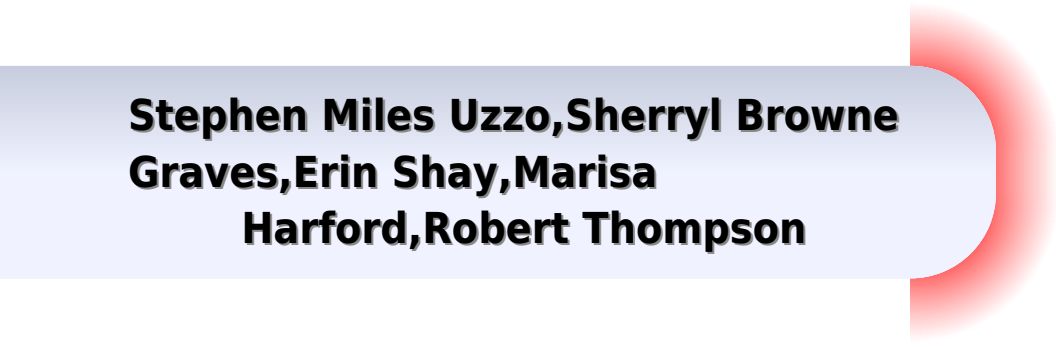


$$Q = m \cdot c \cdot \Delta T(^{\circ}C)$$

$$\Delta T = \frac{Q}{m \cdot c} = \frac{65,000 J}{450g \cdot 4.18 \frac{J}{g^{\circ}C}} = 34.6^{\circ}C = 35^{\circ}C$$

Answer For Quantitative Energy Problems

**Stephen Miles Uzzo, Sherryl Browne
Graves, Erin Shay, Marisa
Harford, Robert Thompson**



Answer For Quantitative Energy Problems:

Quantitative Energy Finance Fred Espen Benth, Valery A. Kholodnyi, Peter Laurence, 2013-08-28 Finance and energy markets have been an active scientific field for some time even though the development and applications of sophisticated quantitative methods in these areas are relatively new and referred to in a broader context as energy finance Energy finance is often viewed as a branch of mathematical finance yet this area continues to provide a rich source of issues that are fuelling new and exciting research developments Based on a special thematic year at the Wolfgang Pauli Institute WPI in Vienna Austria this edited collection features cutting edge research from leading scientists in the fields of energy and commodity finance Topics discussed include modeling and analysis of energy and commodity markets derivatives hedging and pricing and optimal investment strategies and modeling of emerging markets such as power and emissions The book also confronts the challenges one faces in energy markets from a quantitative point of view as well as the recent advances in solving these problems using advanced mathematical statistical and numerical methods By addressing the emerging area of quantitative energy finance this volume will serve as a valuable resource for graduate level students and researchers studying financial mathematics risk management or energy finance

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