

# Matter—Properties and Changes

## Lesson 2.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
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Matter is anything with (1) MASS and volume. A (2) SUBSTANCE is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) PROPERTIES that can be observed. Substances have both physical and chemical properties. (4) PHYSICAL properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) DENSITY are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) CHEMICAL properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

CHEM	7. Chemical formula H <sub>2</sub> O
CHEM	8. Forms green carbonate when exposed to moist air
CHEM	9. Remains unchanged when in the presence of nitrogen
PHYS	10. Colorless
PHYS	11. Solid at normal temperatures and pressures
CHEM	12. Ability to combine with another substance
PHYS	13. Melting point
PHYS	14. Liquid at normal temperatures and pressures
PHYS	15. Boiling point is 100°C
PHYS	16. Conducts electricity

# 3.1 Properties Of Matter Study Guide

**J Spring**

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