

Matter—Properties and Changes

Lesson 2.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical
density

mass
properties

physical
substance

Matter is anything with (1) MASS and volume. A (2) SUBSTANCE is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) PROPERTIES that can be observed. Substances have both physical and chemical properties. (4) PHYSICAL properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) DENSITY are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) CHEMICAL properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

- | | |
|-------------|---|
| <u>CHEM</u> | 7. Chemical formula H_2O |
| <u>CHEM</u> | 8. Forms green carbonate when exposed to moist air |
| <u>CHEM</u> | 9. Remains unchanged when in the presence of nitrogen |
| <u>PHYS</u> | 10. Colorless |
| <u>PHYS</u> | 11. Solid at normal temperatures and pressures |
| <u>CHEM</u> | 12. Ability to combine with another substance |
| <u>PHYS</u> | 13. Melting point |
| <u>PHYS</u> | 14. Liquid at normal temperatures and pressures |
| <u>PHYS</u> | 15. Boiling point is $100^{\circ}C$ |
| <u>PHYS</u> | 16. Conducts electricity |

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