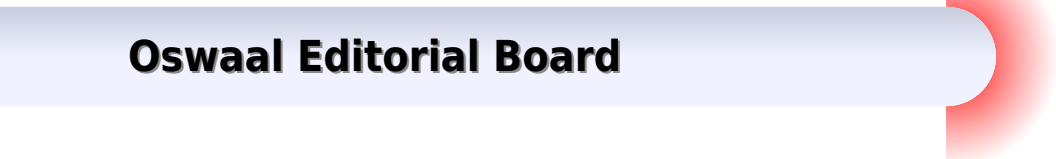


*Bond Energies, D , (kJ/mol)

Single Bonds							
C—H	414	N—H	389	O—H	463	F—F	159
C—C	348	N—N	163	O—O	146	Cl—F	253
C—N	293	N—O	201	O—F	190	Cl—Cl	242
C—O	351	N—F	272	O—Cl	203	Br—F	237
C—F	439	N—Cl	200	O—I	234	Br—Cl	218
C—Cl	328	N—Br	243			Br—Br	193
C—Br	276			S—H	339	I—Cl	298
C—I	238	H—H	436	S—F	327	I—Br	180
C—S	259	H—F	569	S—Cl	251	I—I	151
		H—Cl	431	S—Br	218		
		H—Br	368	S—S	266		
		H—I	297				
Multiple Bonds							
C=C	611	O=O	498				
C≡C	837						
C=N	615	N=N	418				
C≡N	891	N≡N	946				
C=O	799						
C≡O	1072	S=O	523				
		S=S	418				

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