

Chemistry Final Exam Review

Multiple-Choice

Identify the letter of the choice that best completes the statement or answers the question.

1. The study of chemicals that, in general, do not contain carbon is traditionally called what type of chemistry?
a. bio c. physical
b. inorganic d. analytical

2. Which of the following is NOT an example of matter?
a. air c. smoke
b. heat d. water vapor

3. A gold ball has more mass than a tennis ball because it _____.
a. takes up more space c. contains different kinds of matter
b. contains more matter d. has a definite composition

4. An example of an extensive property of matter is _____.
a. temperature c. mass
b. pressure d. hardness

5. All of the following are physical properties of matter EXCEPT _____.
a. mass c. melting point
b. color d. ability to rust

6. Which state of matter has a definite volume and takes the shape of its container?
a. solid c. gas
b. liquid d. both b and c

7. Which of the following CANNOT be classified as a substance?
a. table salt c. nitrogen
b. air d. gold

8. Which of the following is a heterogeneous mixture?
a. air c. steel
b. salt water d. soil

9. Which of the following is a heterogeneous mixture?
a. vinegar in water c. oil and vinegar
b. milk d. air

10. Separating a solid from a liquid by evaporating the liquid is called _____.
a. filtration c. solution
b. condensation d. distillation

11. A substance that can be separated into two or more substances only by a chemical change is an (a) _____.
a. solution c. mixture
b. element d. compound

12. The chemical symbol for iron is _____.
a. fc c. Fe
b. FE d. fr

1question Chemistry Final

**John W. Moore,Conrad L.
Stanitski,Peter C. Jurs**

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