

Acids and Bases

Neutralization Reactions Lab



Reading, Lab, and Practice Bundle (Grades 6-9)

Acids and Bases

What is a Neutralization

A neutralization reaction is when an acid reacts with a base to form a salt and water.

This type of chemical reaction is important in everyday life – for treating an upset stomach or for balancing soil pH in gardens.

What Are Acids and Bases

Acids are substances that taste sour and can have a low pH.

Examples: lemon juice.

Matching Vocabulary

Directions: Match each word on the left with the correct definition. Write the correct letter next to each number.

1. A substance with a pH greater than 7 that produces OH^- ions. ()
2. The product of an acid and base reaction, often found in the form of NaCl . ()
3. A scale from 0 to 14 used to measure how acidic or basic a substance is. ()
4. A chemical that changes color to show if a substance is acidic or basic. ()
5. A substance with a pH less than 7 that produces H^+ ions. ()
6. A reaction between an acid and a base to form water and salt. ()

Matching Vocabulary

NAME: _____

DATE: _____

Directions: Match each word on the left with the correct definition or example on the right. Write the correct letter next to each number.

1. A substance with a pH greater than 7 that produces OH^- ions. ()
2. The product of an acid and base reaction, often found in the form of NaCl . ()
3. A scale from 0 to 14 used to measure how acidic or basic a substance is. ()
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5. A substance with a pH less than 7 that produces H^+ ions. ()
6. A reaction between an acid and a base to form water and salt. ()

Acid

Base

pH scale

Neutralization

Salt

Indicator

28 Pages Printable & Easy to Use Includes Answer Key

19 4 Neutralization Reactions Answer Key

John S. Phillips, Cheryl Wistrom



19 4 Neutralization Reactions Answer Key:

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Fundamentals of Chemistry in the Laboratory Ralph A. Burns, 1999

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Instructors Resource Manual John McMurry, Robert A. Pribush, Robert Fay, 2000-12-27

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