



TENKASI DISTRICT Standard - 12

Time Allowed: 3.00 Hours

CHEMISTRY

Maximum Marks: 70

PART - I

Note: 1. Answer all the questions.**15×1=15****2. Choose the most appropriate answer from the given four alternatives and write the option code and the corresponding answer.**

1. The metal oxide which cannot be reduced to metal by carbon is _____.
 a) PbO b) Al_2O_3 c) ZnO d) FeO
2. Match items in column - I with the items of column - II and assign the correct code.

Column - I

- A. Haematite
 B. Azurite
 C. Galena
 D. Calamine

Column - II

- (i) $2CuCO_3 \cdot Cu(OH)_2$
 (ii) $ZnCO_3$
 (iii) Fe_2O_3
 (iv) PbS

	A	B	C	D
a)	(iii)	(iv)	(i)	(ii)
b)	(iii)	(ii)	(i)	(iv)
c)	(iii)	(iv)	(ii)	(i)
d)	(iii)	(i)	(iv)	(ii)

3. Hybridisation of carbon in Graphite _____.

a) sp^2 b) sp^3 c) sp^2d d) sp

4. Oxidation number of nitrogen in pernitrous acid ($HOONO$) _____.

a) +3 b) +5 c) +1 d) +6

5. Inter halogen compounds of type AX_2 , about the structure of hybridization and bond pairs / lone pairs, which one is correct?

a) T - shaped, sp^3d , $\frac{3}{2}$

b) Square pyramidal, sp^3d^2 , $\frac{5}{1}$

c) T - shaped, sp^3d , $\frac{5}{1}$

d) Square pyramidal, sp^3d , $\frac{3}{2}$

6. Magnetic moment of Mn^{2+} is _____.

a) 5.92 BM b) 8.95 BM c) 2.80 BM d) 3.90 BM

7. Number of electrons transferred to $KMnO_4$ in acid medium _____.

a) 3 b) 1 c) 2 d) 5

8. An ionic compound A_xB_y crystallizes in fcc type crystal structure with B ions at the centre of each face and A ion occupying corners of the cube, the correct formula of A_xB_y is _____.

a) AB b) AB_3
 c) A_3B d) A_8B_6

9. For the reaction, $2SO_2 + O_2 \rightarrow 2SO_3$, if $\frac{-d}{dt}[SO_2] = k_1[SO_2]$, $\frac{-d}{dt}[O_2] = k_2[SO_2]$,

$\frac{-d}{dt}[SO_3] = k_3[SO_2]$ then the relation between k_1 , k_2 and k_3 is _____.

a) $2k_1 = k_2 = k_3$ b) $k_1 = 2k_2 = k_3$

c) $k_1 = k_2 = 2k_3$ d) $2k_1 = 2k_2 = k_3$

10. Which of the following can act as Lowry. Bronsted acid as well as base?

a) HCl b) SO_4^{2-} c) HPO_4^{2-} d) Br

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Joacim Rocklöv

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Table of Contents **12th Chemistry Practice Paper Doc**

1. Understanding the eBook 12th Chemistry Practice Paper Doc
 - The Rise of Digital Reading 12th Chemistry Practice Paper Doc
 - Advantages of eBooks Over Traditional Books
2. Identifying 12th Chemistry Practice Paper Doc
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 12th Chemistry Practice Paper Doc
 - User-Friendly Interface
4. Exploring eBook Recommendations from 12th Chemistry Practice Paper Doc
 - Personalized Recommendations
 - 12th Chemistry Practice Paper Doc User Reviews and Ratings
 - 12th Chemistry Practice Paper Doc and Bestseller Lists

5. Accessing 12th Chemistry Practice Paper Doc Free and Paid eBooks
 - 12th Chemistry Practice Paper Doc Public Domain eBooks
 - 12th Chemistry Practice Paper Doc eBook Subscription Services
 - 12th Chemistry Practice Paper Doc Budget-Friendly Options
6. Navigating 12th Chemistry Practice Paper Doc eBook Formats
 - ePub, PDF, MOBI, and More
 - 12th Chemistry Practice Paper Doc Compatibility with Devices
 - 12th Chemistry Practice Paper Doc Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 12th Chemistry Practice Paper Doc
 - Highlighting and Note-Taking 12th Chemistry Practice Paper Doc
 - Interactive Elements 12th Chemistry Practice Paper Doc
8. Staying Engaged with 12th Chemistry Practice Paper Doc
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 12th Chemistry Practice Paper Doc
9. Balancing eBooks and Physical Books 12th Chemistry Practice Paper Doc
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 12th Chemistry Practice Paper Doc
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 12th Chemistry Practice Paper Doc
 - Setting Reading Goals 12th Chemistry Practice Paper Doc
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 12th Chemistry Practice Paper Doc
 - Fact-Checking eBook Content of 12th Chemistry Practice Paper Doc
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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