

Modern Physics

① Photoelectric Effect :- [PEE]

1. Planck Quantum theory:

$$Q = NE$$

N = No. of Photon's

N = 1, 2, 3, 4, 5, ...

E = Energy of photon.

→ The light energy of a any source is always an integral multiple of a smaller energy [Energy of photon] i.e. Quantum of light.

→ The packets of energy in electromagnetic wave is k/a Photon / Quanta.

→ Acc^d to plank. Energy of photon is directly proportional to the frequency of radiation.

$$E \propto \nu \quad E = h\nu = \frac{hc}{\lambda}$$

$$E(\text{eV}) = \frac{12400}{\lambda(\text{Å})} \quad \text{eV-Å}^{\circ}$$

ν, n, f = frequency

h = Planck's const.

$$h = 6.63 \times 10^{-34} \text{ J-sec.}$$

$$\nu = \frac{c}{\lambda} \quad \text{— speed of light. } (3 \times 10^8 \text{ m/sec})$$

λ — wavelength

$$hc = (6.63 \times 10^{-34}) (3 \times 10^8 \text{ m/sec})$$

$$= 20 \times 10^{-26} = 2 \times 10^{-25} \text{ J-m}$$

$$hc = 12400 \text{ eV-Å}^{\circ}$$

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mathematics physics and statistics biostatistics programs Chapter IX includes a summary of results correlations among measures several additional analyses and suggestions for future studies Among the findings reported are those indicating that mathematics programs had on the average the largest number of faculty N 33 in December 1980 followed closely by physics N 28 and chemistry N 23 and that 80 percent of computer science students had job commitments by graduation Survey instruments and supporting documentation are included in appendices JN

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