

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B,

Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

Gas constant $R = 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$

$= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}$

$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J}$, $1 \text{ Faraday} = 96500 \text{ coulombs}$.

Avogadro's number $= 6.023 \times 10^{23}$.

SECTION A – 14 MARKS

Question 1

- (A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $\text{K}_4[\text{Fe}(\text{CN})_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

Turn over

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Table of Contents 12 Question Paper For Chemistry

1. Understanding the eBook 12 Question Paper For Chemistry
 - The Rise of Digital Reading 12 Question Paper For Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying 12 Question Paper For Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 12 Question Paper For Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from 12 Question Paper For Chemistry
 - Personalized Recommendations
 - 12 Question Paper For Chemistry User Reviews and Ratings
 - 12 Question Paper For Chemistry and Bestseller Lists
5. Accessing 12 Question Paper For Chemistry Free and Paid eBooks

- 12 Question Paper For Chemistry Public Domain eBooks
- 12 Question Paper For Chemistry eBook Subscription Services
- 12 Question Paper For Chemistry Budget-Friendly Options
- 6. Navigating 12 Question Paper For Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - 12 Question Paper For Chemistry Compatibility with Devices
 - 12 Question Paper For Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 12 Question Paper For Chemistry
 - Highlighting and Note-Taking 12 Question Paper For Chemistry
 - Interactive Elements 12 Question Paper For Chemistry
- 8. Staying Engaged with 12 Question Paper For Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 12 Question Paper For Chemistry
- 9. Balancing eBooks and Physical Books 12 Question Paper For Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 12 Question Paper For Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 12 Question Paper For Chemistry
 - Setting Reading Goals 12 Question Paper For Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 12 Question Paper For Chemistry
 - Fact-Checking eBook Content of 12 Question Paper For Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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