

Second Terminal Examination (2022-23)
Subject- CHEMISTRY

Time: - 3 Hrs.

Class- 11th Science

Markas: - 70

Instructions:-

Question paper consist of 31 questions divided into four sections namely A, B, C, D.

- 1) **Section-A:** Q. No.1 contains ten multiple choice types of questions carrying one mark each.
Q.No.2 contains eight very short answer types of questions carrying one mark each.
- 2) **Section-B:** Q.No.3 to Q. No.14 are Twelve questions carrying two mark each out of which any eight to be attempted.
- 3) **Section-C:** Q. No.15 to Q.No.26 are Twelve questions carrying three mark each out of which any eight are to be attempted.
- 4) **Section-D:** Q.No.27 to Q.No. 31 are Five questions carrying four mark each out of which any three to be attempted.

Section- A

Q.1) Select and write the correct answer:

(10)

- i. Which of the following compounds can NOT demonstrate the law of proportions?
a) NO_2NO_2 b) CO_2CO_2 c) $\text{H}_2\text{O}_2\text{H}_2\text{O}_2$ d) $\text{Na}_2\text{S}_2\text{Na}_2\text{S}_2$
- ii. Which of the following temperature will read the same value on Celsius and Fahrenheit Scales
a) -40 b) +40 c) -80 d) -20
- iii. SI unit of the quantity electric current is
a) Volts b) Ampere c) Candela d) Newton
- iv. The no. of molecules in 22.4 cm³ of Nitrogen Gas at STP is
a) 6.022×10^{20} b) 6.022×10^{23} c) 22.4×10^{20} d) 22.4×10^{23}
- v. The homogeneous series of alcohol has general molecular formula ?
a) $\text{C}_n\text{H}_{2n+1}\text{OH}$ b) $\text{C}_n\text{H}_{2n}\text{OH}$ c) $\text{C}_n\text{H}_{2n+1}\text{OH}$ d) none
- vi. Find the molar mass of CH_3CHO .
a) 44 b) 40 c) 80 d) 30
- vii. Find the molar mass of CH_3NO_2 .
a) 61 b) 77 c) 98 d) 51
- viii. Atomic number of an elements is equal to
a) $P = e^-$ b) $p = N$ c) $P = A$ d) $p = n = e$

11th Question Paper Chemistry 23

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