

**Q. 3. Answer the following questions.**

- (i) Discuss Law of Reciprocal Proportion with examples.
- (ii) An inorganic salt gave the following percentage composition:  
 $\text{Na} = 29.11\%$  ,  $\text{S} = 40.51\%$  &  $\text{O} = 30.38\%$  , Calculate the Empirical Formula of the salt.
- (iii) Discuss Aufbau's Principle and Hund's Rule.
- (iv) What is diagonal relationship with example.
- (v) Explain  $sp^2$  hybridization with example.
- (vi) On the basis of VSEPR theory, predict the shapes of the following molecules/ ions?

(a) $\text{SF}_4$	(b) $\text{NH}_4^+$	(c) $\text{H}_3\text{O}^+$
(d) $\text{NH}_2$	(e) $\text{PCl}_5$	(f) $\text{NH}_3$
- (vii) State & explain Hess's Law of constant heat summation.
- (viii) What is Buffer Solution. Give its types.
- (ix) Give three methods of preparation of Alkynes.

#### **SECTION D**

**Q. 4. Answer the following questions.**

(i) What are Quantum numbers? Explain

Or

State and Explain Heisenberg's Uncertainty Principle.

(ii) Define 1st Law of Thermodynamics. Derive its mathematical formulation.

Or

Define Heat Capacity. Derive relation between  $C_p$  &  $C_v$ .

(iii) Explain Nucleophilic Substitution Reaction.

Or

Write the IUPAC names of the following organic compounds:



# 11th Question Paper Chemistry 2013

**Lakshman Prasad,Deepak  
paliwal,Mansi Garg,Neha  
Minglani,Sanubia Saleem**

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