

Stoichiometry Worksheet 2

1. Sodium chloride can be produced by the reaction of sodium and chlorine gas. Write a balanced equation for this reaction.

Suppose that 6.70 mol Na react with 3.20 mol Cl_2 .

- What is the limiting reagent?
- How many moles of NaCl are produced?

2. The equation for the complete combustion of ethane, C_2H_6 , is



Balance the equation. If 2.70 mol C_2H_6 is reacted with 6.30 mol O_2 ,

- identify the limiting reagent.
- calculate the moles of water produced.

3. Copper and sulfur combine to form copper (I) sulfide.



- What is the limiting reagent when 80.0 g Cu reacts with 25.0 g S?
- What is the maximum number of grams of Cu_2S that can be formed?

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**Washington University (Saint Louis,
Mo.)**



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A Stoichiometry Unit David Callaghan, 2004 **Jacaranda Chemistry 2 VCE Units 3 and 4, 3e learnON and Print**
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to Chemistry, Laboratory Manual T. R. Dickson, 1994-12-23 Teaches chemistry by offering a dynamic provocative and
relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text
developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and
discussion activities These themes involve and engage the students in their own learning processes they are challenged to be
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thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts
in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the
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