

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the processes of a reaction.
- 2. Explain how to write a balanced equation.
- 3. List the steps for writing a complete ionic equation.

Vocabulary

- 1. chemical equation
- 2. balanced equation
- 3. catalyst
- 4. coefficient
- 5. skeletal equation

Part A Completion

Write the coefficient for each in the balanced equation, and balance the equation by changing the coefficient for each substance. Write balanced ionic equations for each.

1. A chemical equation is always represented by a balanced _____ 2. _____
 The chemical formula for a chemical compound is _____ 3. _____
 Chemical formulas for compounds are the _____ 4. _____
 The law of conservation of _____ 5. _____
 A chemical equation is balanced when the number of _____ 6. _____
 atoms of each element is the same on both sides of the equation. _____ 7. _____
 To balance a chemical equation, you place _____ 8. _____
 coefficients in front of the chemical formulas. _____ 9. _____
 The _____ 10. _____
 _____ 11. _____
 _____ 12. _____
 _____ 13. _____
 _____ 14. _____
 _____ 15. _____
 _____ 16. _____
 _____ 17. _____
 _____ 18. _____
 _____ 19. _____
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_____ 21. _____
 _____ 22. _____
 _____ 23. _____
 _____ 24. _____
 _____ 25. _____
 _____ 26. _____
 _____ 27. _____
 _____ 28. _____
 _____ 29. _____
 _____ 30. _____

- 1. equation
- 2. formula
- 3. chemical
- 4. law
- 5. mass
- 6. balanced
- 7. coefficient
- 8. atoms
- 9. balanced
- 10. balanced
- 11. balanced
- 12. balanced
- 13. balanced
- 14. balanced
- 15. balanced
- 16. balanced
- 17. balanced
- 18. balanced
- 19. balanced
- 20. balanced

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D Keegan

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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Columbia Review High-yield General Chemistry Stephen D. Bresnick, 1996 A study guide in question and answer format for basic chemistry *Prentice Hall Chemistry*, 2000 *Energetic Principles of Chemical Reactions* Jack Simons, 1983
Chemical Engineering and Mining Review, 1955 **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **On Solar Hydrogen and Nanotechnology** Lionel Vayssieres, 2010-01-26 More energy from the sun strikes Earth in an hour than is consumed by humans in an entire year Efficiently harnessing solar power for sustainable generation of hydrogen requires low cost purpose built functional materials combined with inexpensive large scale manufacturing methods These issues are comprehensively addressed in *On Solar Hydrogen Nanotechnology* an authoritative interdisciplinary source of fundamental and applied knowledge in all areas related to solar hydrogen Written by leading experts the book emphasizes state of the art materials and characterization techniques as well as the impact of nanotechnology on this cutting edge field Addresses the current status and prospects of solar hydrogen including major achievements performance benchmarks technological limitations and crucial remaining challenges Covers the latest advances in fundamental understanding and development in photocatalytic reactions semiconductor nanostructures and heterostructures quantum confinement effects device fabrication modeling simulation and characterization techniques as they pertain to solar generation of hydrogen Assesses and establishes the present and future role of solar hydrogen in the hydrogen economy Contains numerous graphics to illustrate concepts techniques and research results *On Solar Hydrogen Nanotechnology* is an essential reference for materials scientists physical and inorganic chemists electrochemists physicists and engineers carrying out research on solar energy photocatalysis or semiconducting nanomaterials both in academia and industry It is also an invaluable resource for graduate students and postdoctoral researchers as well as business professionals and consultants with an interest in renewable energy *Modern Chemistry* Nicholas D. Tzimopoulos, 1993
The Lancet, 1895 *Chemistry* Henry Dorin, 1987 **Official Gazette of the United States Patent Office** USA Patent Office, 1880 **Russian Chemical Reviews**, 1994 *Battelle Technical Review*, 1954 **The Science Teachers Bulletin**, 1971 Bibliography of Scientific and Industrial Reports, 1966-03 **Corrosion Abstracts**, 1969 Anderson's Essentials of Biochemistry Gordon Herman Pritham, 1968

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