

## Angle of Elevation & Depression Trig Worksheet

1. Brian's kite is flying above a field at the end of 65 m of string. If the angle of elevation to the kite measures  $70^\circ$ , how high is the kite above Brian's head?



$$\sin 70^\circ \frac{x}{65}$$

$$x \approx 61.06 \text{ m}$$

2. From an airplane at an altitude of 1200 m, the angle of depression to a building on the ground measures  $28^\circ$ . Find the distance from the plane to the building.



$$\sin 28^\circ \frac{1200}{x}$$

$$x = \frac{1200}{\sin 28^\circ} \rightarrow 2336.07 \text{ m}$$

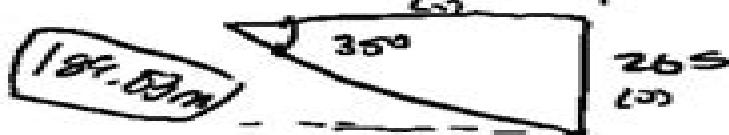
3. From a point on the ground 12 ft from the base of a flagpole, the angle of elevation of the top of the pole measures  $53^\circ$ . How tall is the flagpole?



$$\tan 53^\circ \frac{x}{12}$$

$$x \approx 15.92 \text{ ft}$$

4. From a plane flying due east at 265 m above sea level, the angles of depression of two ships sailing due east measure  $35^\circ$  and  $25^\circ$ . How far apart are the ships?



$$\tan 35^\circ \frac{x}{a}$$

$$x = \frac{265}{\tan 35^\circ} \approx 278.46$$

$$\tan 25^\circ \frac{x}{a} = \frac{265}{\tan 25^\circ} = 566.20$$

5. A man flies a kite and lets out 100 feet of string. The angle of elevation of the string is  $52^\circ$ . How high off the ground is the kite? How far away is the man from the spot directly under the kite?



$$\sin 52^\circ \frac{x}{100}$$

$$x \approx 78.8 \text{ ft}$$

$$\cos 52^\circ \frac{x}{100}$$

$$x \approx 61.5 \text{ ft}$$

6. From the top of a vertical cliff 40 m high, the angle of depression of an object that is level with the base of the cliff is  $34^\circ$ . How far is the object from the base of the cliff?



$$\tan 34^\circ: \frac{40}{x}$$

$$x = \frac{40}{\tan 34^\circ}$$

$$x \approx 59.3 \text{ m}$$

# Practice 8 4 Angles Of Elevation And Depression

## Answers

**S Baum**

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2  $\text{CO}_2 \text{ g} \rightarrow \text{H}_2 \text{O g}$   $\Delta h = 57 \text{ kJ}$   
2  $\text{H}_2 \text{ g} \rightarrow \text{H}_2 \text{O g}$   $\Delta h = 94.5 \text{ kJ}$   
 $\text{C}_2\text{H}_6 \text{ g} \rightarrow \text{H}_2 \text{ g} + \text{C}_2\text{H}_5 \text{ g}$   $\Delta h = 71.2 \text{ kJ}$   
2  $\text{O}_2 \text{ g} \rightarrow 3 \text{ H}_2 \text{O g}$   $\Delta h = 283 \text{ kJ}$   
 $\text{C}_2\text{H}_6 \text{ g} \rightarrow \text{C}_2\text{H}_5 \text{ g} + \text{H}_2 \text{ g}$   $\Delta h = 1560 \text{ kJ}$

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3  $\text{O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g}$   $\Delta h = 1411 \text{ kJ}$  mole  
2  $\text{H}_2 \text{ g} \rightarrow 2 \text{ O}_2 \text{ g}$   $\Delta h = 1560 \text{ kJ}$  mole  
 $\text{C}_2\text{H}_4 \text{ g} \rightarrow \text{C}_2\text{H}_5 \text{ g} + \text{H}_2 \text{ g}$   $\Delta h = 285 \text{ kJ}$  mole

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values to determine the heat of reaction for the reaction c 2 h 4

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web thus for reaction 9 2 in which 1 mole of h<sub>3</sub>o and 1 mole of oh combine to form 2 moles of h<sub>2</sub>o h 55 8 kj as shown note that the enthalpy change h hproducts hreactants is positive if heat is absorbed that is if hproducts

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