

**Atomic Theory Practice Test**

Name:

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- \_\_\_\_\_ 1. If two atoms of Ca have a different number of electrons, which property (or properties) would be significantly different?
- Mass
  - Charge
  - Both A & B
  - Neither A nor B
- \_\_\_\_\_ 2. If two atoms of Na have a different number of neutrons which property (or properties) would be significantly different?
- Mass
  - Charge
  - Both A & B
  - Neither A nor B
- \_\_\_\_\_ 3. Which of the following ions will NOT have the same electron configuration as Ne?
- $\text{Na}^+$
  - $\text{Al}^{3+}$
  - Ar
  - $\text{O}^{2-}$
- \_\_\_\_\_ 4. Which of the following easily loses one outermost electron?
- Potassium
  - Beryllium
  - Iron
  - Bromine
- \_\_\_\_\_ 5. Which of the following would have the largest atomic radius?
- Seaborgium
  - Einsteinium
  - Indium
  - Thallium
- \_\_\_\_\_ 6. Which trend in the halogen family occurs with increasing atomic number?
- Ionization energies decrease
  - Atomic radii decrease
  - Electronegativities increase
  - Tendency to gain electrons increases
- \_\_\_\_\_ 7. Which of the following species would have 2 valence electrons?
- Boron
  - Fluorine
  - Oxygen ion
  - Vanadium (III) ion
- \_\_\_\_\_ 8. What sub-shell is especially stable when it is half-filled?
- s-subshell
  - p-subshell
  - d-subshell
  - f-subshell

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**O García**



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  - Chapter 1: Introduction to Atomic Theory Practice Test
  - Chapter 2: Essential Elements of Atomic Theory Practice Test
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