

Fill in the blank (2 points each blank)

- Elements that have the properties of both metal and non-metals are metalloids.
- What do elements in the same group have the same number of? valence electrons.
- Which element would have the highest electron affinity? Fluorine.
- What element is at period 3, group 8A? Argon.
- How many valence electrons does Bromine have? 7.
- The atomic number tells you how many protons (and electrons if neutral) there are in an atom.
- Two Characteristics of Non-metals are not good conductors of heat or electricity, not malleable, etc.
- Two characteristics of Metals are malleable, ductile, good conductors of heat and electricity.
- Which element would have the smallest atomic radius? Helium.
- Which element would have the highest ionization energy? Helium.

11. For the questions below and based on the information provided, fill in the missing information. (1 point each)

Name	Symbol	# of protons	# of electrons	# of neutrons	Atomic Number	Mass Number
Potassium	K	19	19	22	19	41
Phosphorous	P	15	15	18	15	33
Beryllium	Be	4	4	4	4	8
Sulfur	S	16	16	20	16	36

12. Draw Bohr Models for (3 points each)

a. Na^{2+} b. N^{15}

13. What is an isotope? (1 point)

An atom with a different number of neutrons

14. How can you tell the identity of any atom? (1 point)

Number of PROTONS

15. Allen Periodic Chart -- Directions: Read the clue and place the alien symbol in the corresponding box!

Atoms And Periodic Table Test Answer Key

Ralph A. Burns, John William Hill

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