

Name: _____

Period: _____

Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:

number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



Protons:

7

Electrons:

7

Neutrons:

7

Protons:

1

Electrons:

1

Neutrons:

0

Protons:

8

Electrons:

8

Neutrons:

8

BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
 1. The first energy level can hold 2 electrons
 2. The second energy level can hold 8 electrons
 3. The third energy level can hold 8 electrons

The electrons in the outermost energy level are called:

Valence Electrons

Element: SulfurNumber of Valence Electrons: 6

LEWIS DIAGRAMS

A lewis diagram shows only the symbol with the Valence electrons



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RC Schank



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1. Understanding the eBook Atomic Structure Worksheet Group Document Sample
 - The Rise of Digital Reading Atomic Structure Worksheet Group Document Sample
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Worksheet Group Document Sample
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Worksheet Group Document Sample
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Worksheet Group Document Sample
 - Personalized Recommendations
 - Atomic Structure Worksheet Group Document Sample User Reviews and Ratings
 - Atomic Structure Worksheet Group Document Sample and Bestseller Lists
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Worksheet Group Document Sample
 9. Balancing eBooks and Physical Books Atomic Structure Worksheet Group Document Sample
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Worksheet Group Document Sample
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Atomic Structure Worksheet Group Document Sample
 - Setting Reading Goals Atomic Structure Worksheet Group Document Sample
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Atomic Structure Worksheet Group Document Sample
 - Fact-Checking eBook Content of Atomic Structure Worksheet Group Document Sample
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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