

CLASS:XI CHEMISTRY
CHAPTER#02
ATOMIC STRUCTURE
BY: SIR AAMIR HULLIO

1. Cathode rays are deflected by
(a) electric field only (b) magnetic field only (c) electric and magnetic field (d) none of these
2. The e/m value for the particles constituting cathode rays is the same regardless of
(a) the gas present in cathode rays tube (b) the metal of which cathode was made
(c) both of these (d) none of these
3. The charge to mass ratio (e/m) of positive particles
(a) varies with the nature of gas in discharge tube (b) is independent of the gas in discharge tube
(c) is constant (d) none of the above
4. A sub atomic particle which has one unit mass and one unit positive charge is known as
(a) hydrogen atom (b) neutron (c) electron (d) proton
5. Atomic number of an element is equal to the number of _____ in the nucleus of the atom.
(a) neutrons (b) protons (c) both the neutrons and protons (d) electrons
6. The mass number of an atom is equal to the number of _____ in the nucleus of an atom
(a) protons (b) neutrons (c) electrons (d) nucleons
7. The mass number and atomic number of Phosphorus atom are 31 and 15 respectively. The number of neutrons in the nucleus is
(a) 15 (b) 16 (c) 31 (d) 46
8. In a sodium atom (atomic number = 11 and mass number = 23), the number of neutrons is
(a) equal to the number of protons (b) less than the number of protons
(c) greater than the number of protons (d) none of these
9. Which of the following is not correct for electromagnetic waves?
(a) the wavelength is the distance between two successive crests
(b) the frequency is the number of waves which pass a given point in one second
(c) the velocity of a wave is the distance covered by the particular wave in one second
(d) all electromagnetic waves have equal wavelengths
10. If Z is the number of proton and A the number of nucleons, then the number of neutrons in an atom is given by
(a) $A + Z$ (b) $A - Z$ (c) $Z - A$ (d) none of these
11. The mass number and atomic number of Phosphorus atom are 31 and 15 respectively. The number of neutrons in the nucleus is
(a) 15 (b) 16 (c) 31 (d) 46
12. Which of the following relations is not correct?
(a) $c = \lambda \times \mu$ (b) $\mu = 1/f$ (c) $\mu = 1/\lambda$ (d) $\lambda = c + \mu$
13. The unit in which wave number is measured
(a) hertz (b) sec^{-1} (c) nanometer (d) cm^{-1}
14. In the spectrum of hydrogen atom, the series which falls in ultraviolet region is
(a) Lyman series (b) Balmer series (c) Paschen series (d) Brackett series
15. The Balmer series in the spectrum of hydrogen atom falls in
(a) ultraviolet region (b) visible region (c) infrared region (d) none of these
16. The radius of first orbit in hydrogen atom is 0.529 \AA . The radius of second orbit is given by
(a) $\frac{1}{2} \times 0.529 \text{ \AA}$ (b) $2 \times 0.529 \text{ \AA}$ (c) $4 \times 0.529 \text{ \AA}$ (d) $8 \times 0.529 \text{ \AA}$

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Table of Contents Atomic Structure Ch 4 Packet Answers

1. Understanding the eBook Atomic Structure Ch 4 Packet Answers
 - The Rise of Digital Reading Atomic Structure Ch 4 Packet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Ch 4 Packet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Ch 4 Packet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Ch 4 Packet Answers
 - Personalized Recommendations
 - Atomic Structure Ch 4 Packet Answers User Reviews and Ratings
 - Atomic Structure Ch 4 Packet Answers and Bestseller Lists
5. Accessing Atomic Structure Ch 4 Packet Answers Free and Paid eBooks
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- Atomic Structure Ch 4 Packet Answers Budget-Friendly Options

6. Navigating Atomic Structure Ch 4 Packet Answers eBook Formats

- ePub, PDF, MOBI, and More
- Atomic Structure Ch 4 Packet Answers Compatibility with Devices
- Atomic Structure Ch 4 Packet Answers Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Atomic Structure Ch 4 Packet Answers
- Highlighting and Note-Taking Atomic Structure Ch 4 Packet Answers
- Interactive Elements Atomic Structure Ch 4 Packet Answers

8. Staying Engaged with Atomic Structure Ch 4 Packet Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Atomic Structure Ch 4 Packet Answers

9. Balancing eBooks and Physical Books Atomic Structure Ch 4 Packet Answers

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Atomic Structure Ch 4 Packet Answers

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Atomic Structure Ch 4 Packet Answers

- Setting Reading Goals Atomic Structure Ch 4 Packet Answers
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Atomic Structure Ch 4 Packet Answers

- Fact-Checking eBook Content of Atomic Structure Ch 4 Packet Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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