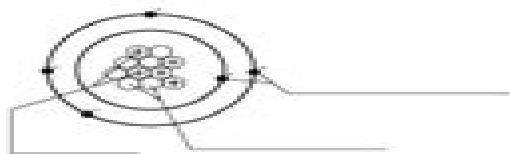


Basic Atomic Structure

1. Label the parts of an atom in the diagram below



- What type of charge does a proton have? _____
- What type of charge does a neutron have? _____
- What type of charge does an electron have? _____
- Which two subatomic particles are located in the nucleus of an atom? _____

2. An element is represented by its chemical symbol along with a few numbers. Based on the example on the right, fill in the numbers of protons (P), neutrons (N), and electrons (E) for the following elements.

8	← Atomic Number
O	← Symbol
Oxygen	← Name
15.999	← Average atomic mass

6	#P _____
C	#N _____
Carbon	#E _____
12.011	

10	#P _____
Ne	#N _____
Neon	#E _____
20.108	

19	#P _____
K	#N _____
Potassium	#E _____
39.098	

3. Complete the table using your knowledge of the periodic table.

Symbol	Atomic Number	Mass Number	Number of Protons	Number of Electrons	Number of Neutrons
Na			11		12
K		39		19	
			38		50
F				9	10
	20	40		20	
	50			50	69
I	53	127			

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Exploration of the Structure of Atom Chandan Sengupta, First Publication April 2019 Revised Publication October 2022 Second Revised Edition July 2023 Third Revised Edition December 2024 Published by Chandan Sengupta Printed by IECIT Printing and Publication Services Resource Centre Arabinda Nagar Bankura 722101 W B India Topics Covered Atoms and Molecules Structure of Atom Bonding Mechanism and Chemical Reactions Mechanism of Bonding This book is suitable for students of Class 9 to Class 11 Students aspiring for Pre Medical Entrance Examination can also get adequate support Additional Hard Copies can also be obtained from Chandan Sukumar Sengupta Arabinda Nagar Bankura 722101 WB Write to Us for more materials

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- 8 Extraction Of Metal The S Block Elements
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- 10 Hydrocarbons Halogen Derivatives
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Table of Contents Atomic Structure 33 Answers

1. Understanding the eBook Atomic Structure 33 Answers
 - The Rise of Digital Reading Atomic Structure 33 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure 33 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure 33 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure 33 Answers
 - Personalized Recommendations
 - Atomic Structure 33 Answers User Reviews and Ratings
 - Atomic Structure 33 Answers and Bestseller Lists
5. Accessing Atomic Structure 33 Answers Free and Paid eBooks
 - Atomic Structure 33 Answers Public Domain eBooks
 - Atomic Structure 33 Answers eBook Subscription Services

- Atomic Structure 33 Answers Budget-Friendly Options
- 6. Navigating Atomic Structure 33 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure 33 Answers Compatibility with Devices
 - Atomic Structure 33 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure 33 Answers
 - Highlighting and Note-Taking Atomic Structure 33 Answers
 - Interactive Elements Atomic Structure 33 Answers
- 8. Staying Engaged with Atomic Structure 33 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure 33 Answers
- 9. Balancing eBooks and Physical Books Atomic Structure 33 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure 33 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure 33 Answers
 - Setting Reading Goals Atomic Structure 33 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure 33 Answers
 - Fact-Checking eBook Content of Atomic Structure 33 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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