

Name: _____

Period: _____

Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:

number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



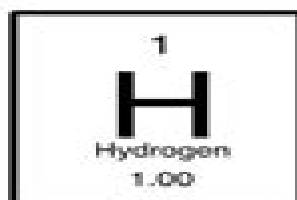
Protons:

7

Electrons:

7

Neutrons:

7

Protons:

1

Electrons:

1

Neutrons:

0

Protons:

8

Electrons:

8

Neutrons:

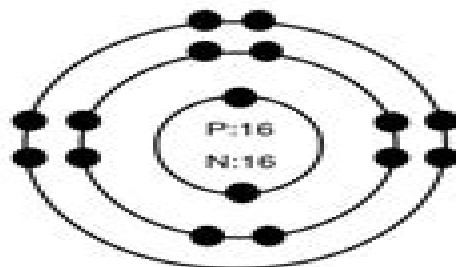
8

BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
 1. The first energy level can hold 2 electrons
 2. The second energy level can hold 8 electrons
 3. The third energy level can hold 8 electrons

The electrons in the outermost energy level are called:

Valence Electrons

Element: SulfurNumber of Valence Electrons: 6

LEWIS DIAGRAMS

A lewis diagram shows only the symbol with the Valence electrons



Atomic Basics Atomic Structure Worksheet Answers

Franklin S. Allaire, Jennifer E. Killham



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Teaching and Learning Online Franklin S. Allaire, Jennifer E. Killham, 2023-01-10 Teaching and Learning Online Science for Elementary Grade Levels explores the challenges of teaching science virtually It includes sections on frameworks teacher journeys and lesson plans aligned with Next Generation Science Standards offering tips resources and discussion questions for educators and students

Learning Elementary Chemistry for Class 8 (A.Y. 2023-24) Onward Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum The content of this series has been developed to fulfill the requirement of all the six domains Concepts Processes Applications Attitudes Creativity and World view of Science to make teaching and learning of Chemistry interesting understandable and enjoyable for young minds This series builds a solid foundation for young learners to prepare them for higher classes The main strength of the series lies in the subject matter and the experience that a learner will get in solving difficult and complex problems of Chemistry Emphasis has been laid upon mastering the fundamental principles of Chemistry rather than specific procedures Unique features of this series are The content of the book is written in a very simple and easy to understand language All the Key concepts in the curriculum have been systematically covered and graded in the text Each theme has been divided into units followed by thought provoking and engaging exercises to test the knowledge understanding and applications of the concepts learnt in that unit At the end of each theme a comprehensive theme assignment which is aligned with the guidelines provided in National Education Policy NEP 2020 is given Explanations illustrations diagrams experiments and solutions to numerical problems have been included to make the subject more interesting comprehensive and appealing Diagrams illustrations and text have been integrated to enhance comprehension Definitions and other important scientific information are highlighted Throughout the series investigations related to the text enable the learners to learn through experimentation Quick revision of each chapter has been given under the caption Highlights in Review Online Support It provides Video lectures Unit wise interactive exercises Chapterwise Worksheet Solution of textbook questions for Teachers only E Book for Teachers only I hope this series would meet the needs and requirements of the curriculum to achieve the learning outcomes as laid down in the curriculum Suggestions and constructive feedback for the further improvement of the book shall be gratefully acknowledged and incorporated in the future edition of the book Author

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 - The Rise of Digital Reading Atomic Basics Atomic Structure Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Basics Atomic Structure Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Basics Atomic Structure Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Basics Atomic Structure Worksheet Answers

- Personalized Recommendations
 - Atomic Basics Atomic Structure Worksheet Answers User Reviews and Ratings
 - Atomic Basics Atomic Structure Worksheet Answers and Bestseller Lists
5. Accessing Atomic Basics Atomic Structure Worksheet Answers Free and Paid eBooks
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 - Atomic Basics Atomic Structure Worksheet Answers eBook Subscription Services
 - Atomic Basics Atomic Structure Worksheet Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Atomic Basics Atomic Structure Worksheet Answers Compatibility with Devices
 - Atomic Basics Atomic Structure Worksheet Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Basics Atomic Structure Worksheet Answers
 - Highlighting and Note-Taking Atomic Basics Atomic Structure Worksheet Answers
 - Interactive Elements Atomic Basics Atomic Structure Worksheet Answers
 8. Staying Engaged with Atomic Basics Atomic Structure Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Basics Atomic Structure Worksheet Answers
 9. Balancing eBooks and Physical Books Atomic Basics Atomic Structure Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Basics Atomic Structure Worksheet Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Atomic Basics Atomic Structure Worksheet Answers
 - Setting Reading Goals Atomic Basics Atomic Structure Worksheet Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Atomic Basics Atomic Structure Worksheet Answers

- Fact-Checking eBook Content of Atomic Basics Atomic Structure Worksheet Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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