

13. (4 pts) What is an ionic compound?
a. A compound that will dissolve in water
b. Two nonmetal atoms bonded together
c. Two metal atoms bonded together
d. A metal and a nonmetal atom bonded together
e. None of the above

14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
a. Ionic
b. Covalent
c. Both a and b
d. Neither a nor b

15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
a. Ionic
b. Covalent
c. Both a and b
d. Neither a nor b

16. (4 pts) How many total atoms are in this molecule: $\text{C}_6\text{H}_{12}\text{OH}$?
a. 3
b. 4
c. 19
d. 20
e. None of the above

17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is $\text{C}_x\text{H}_{10}\text{Cl}_3$, what must x be?
a. 1
b. 2
c. 3
d. 4
e. None of the above

18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water? $\text{C}_6\text{H}_5\text{ClF}$ $\text{C}_6\text{H}_5\text{CH}_3$ CS_2 BaSO_4
a. $\text{C}_6\text{H}_5\text{ClF}$
b. $\text{C}_6\text{H}_5\text{CH}_3$
c. CS_2
d. BaSO_4
e. CS_2 and BaSO_4

Answers For Chemistry Test Module 13

Bert Spilker

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