

Chemistry 113 Practice Exam 4 Spring 2016

Directions on actual Exam: The exam consists of 2 sections (Chemical Bonding and Geometry and IMFs. Each multiple choice question is worth 4.33 points. You are allowed to use a scientific (nonprogrammable) calculator, periodic table, and VSEPR Theory Table.

Useful information:

$$\Delta H^{\circ}_{\text{rxn}} = \sum n \times D(\text{bonds broken}) - \sum n \times D(\text{bonds formed})$$

where $\sum n$ represents the sum of the moles of a particular type of bond

D represents bond energy per mole of bond

Chemical Bonding and Geometry

1.

The Lewis dot symbol for the calcium ion is

A)

2+

B)

$\square \text{Ca} \square$

C)

2+

D)

Ca^{2+}

E)

Ca

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Graham Bone



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