

Study Guide: Alkenes, Cycloalkenes, Alkynes, and Cycloalkynes

Reactions of Alkenes and Cycloalkenes

General Reactivity

- Alkenes and cycloalkenes are unsaturated hydrocarbons characterized by at least one carbon–carbon double bond ($C=C$).
- Their most common reactions are electrophilic additions, where the π bond breaks to form two new σ bonds.
- Cycloalkenes behave similarly to alkenes but their reactivity is influenced by ring strain and the stability of carbocations formed during reactions.

Common Addition Reactions

1. Hydrogenation

Definition: Addition of hydrogen (H_2) across a double bond.

Catalyst: Ni, Pt, or Pd.

Product: Converts alkene \rightarrow alkane.

Example: Ethene + H_2 \rightarrow Ethane.

2. Halogenation

Definition: Addition of a halogen (Cl_2 or Br_2) across the $C=C$ bond.

Product: Dihaloalkane.

Test: Decolorization of bromine water indicates presence of a double bond.

Alkenes And Alkynes Study Guide Answer Key

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