

# Electron Configuration (Ions / Valence) - Worksheet #6

Symbol	# of Electrons	Electron Configuration (Use EITHER method, Core or "Long")
$Ca^{+2}$	18	$Ca^{+2} : [Ar]^{18}$
$Se^{-2}$	32	$Se^{-2} : [Ar]^{18} 3s^2 3p^6$
$Sn^{+2}$	22	$Sn^{+2} : [Kr]^{18} 5s^2 4s^2$
$Zn^{+2}$	10	$Zn^{+2} : [Ar]^{10} 3d^10$
$Sr^{+2}$		$Sr^{+2} : [Kr]^{18}$
$O^{-2}$	6	$O^{-2} : [He]^{2} 2s^2 2p^4$
$Cs^{+1}$		$Cs^{+1} : [Xe]^{18}$
$I^{-}$	8	$I^{-} : [Kr]^{18} 5s^2 4s^2 3p^6$
$Fe^{+3}$	7	$Fe^{+3} : [Ar]^{14} 4s^1 3d^5$
$Cu^{+2}$	9	$Cu^{+2} : [Ar]^{10} 4s^2 3d^9$

Directions: State how many valence electrons

1. O \_\_\_\_\_ 6 \_\_\_\_\_

2. P \_\_\_\_\_ 5 \_\_\_\_\_

3. V \_\_\_\_\_

4. Ca \_\_\_\_\_ 2 \_\_\_\_\_

5. Xe \_\_\_\_\_ 0 \_\_\_\_\_

6. Hg \_\_\_\_\_

7. Te \_\_\_\_\_ 6 \_\_\_\_\_

8. Cl<sup>-</sup> \_\_\_\_\_

9. I<sup>-3</sup> \_\_\_\_\_

10. Zn<sup>+2</sup> \_\_\_\_\_

11. Ge<sup>-4</sup> \_\_\_\_\_

12. Sb<sup>+3</sup> \_\_\_\_\_

13. O<sup>-</sup> \_\_\_\_\_

14. Tc<sup>-4</sup> \_\_\_\_\_

## 5 2 Electron Configurations Answer Key

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,1975 This report covers the year 1972 and lists approximately 10 000 articles of interest to mass spectroscopists This two volume report consists of three sections Vol II contains the Key Word Out of Context Index KWOC Index section The KWOC Index lists the key words the reference numbers of the articles in which the key word appears and the first 100 characters of the title [General Chemistry](#) Kenneth W. Whitten,Kenneth D. Gailey,Raymond E. Davis,1988

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Martin Stuart Silberberg,1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives [Fusion Technology](#) ,1989

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