

Multiple Choice Questions for AQA AS Chemistry

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AQA A Level Chemistry Physical Chemistry 1

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Amount of Substance

This test consists of **TEN** multiple choice questions. You should choose **ONE** answer for each question.

1. How many molecules are present in 15.0 g of PH_3 ?
(Avogadro's number is 6.022×10^{23})

- ☐ A 2.66×10^{24}
☐ B 2.55×10^{24}
☐ C 1.36×10^{24}
☐ D 1.37×10^{24}

2. What is the mass number of oxygen?

- ☐ A 8
☐ B 16
☐ C 32
☐ D 64

3. The density of pure water is 1000 g dm^{-3} . What is the concentration of pure water in mol dm^{-3} ?

- ☐ A 1 mol dm^{-3}
☐ B 56 mol dm^{-3}
☐ C 32 mol dm^{-3}
☐ D 18 mol dm^{-3}

4. Calculate the mass of KOH that should be dissolved in 500 cm^3 water to give a solution with a concentration of 0.6 mol dm^{-3}

- ☐ A 0.0673 g
☐ B 18.7 g
☐ C 0.3 g
☐ D 16.8 g

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5. A 1 dm^3 sample of Noble gas X contains 0.04471 moles. The density of the gas is 1.784 g dm^{-3} . Which gas does the sample contain?

- ☐ A Helium
☐ B Neon
☐ C Argon
☐ D Krypton

6. Calculate the concentration of the solution that is formed by adding 50 cm^3 of water to 25 cm^3 of $0.268 \text{ mol dm}^{-3} \text{ HCl}$

- ☐ A $0.134 \text{ mol dm}^{-3}$
☐ B $0.0893 \text{ mol dm}^{-3}$
☐ C 1.34 mol dm^{-3}
☐ D $3.35 \times 10^{-2} \text{ mol dm}^{-3}$

7. 0.05 moles of an iron salt FeCl_x reacts with AgNO_3 to form 21.5 g of AgCl . What is the value of x ?

- ☐ A 1
☐ B 2
☐ C 3

8. A The uncertainty of a balance is $\pm 0.1 \text{ g}$. What is the % uncertainty if this balance is used to measure 26.7 g of CuSO_4 ?

- ☐ A 0.4%
☐ B 0.1%
☐ C 0.004%
☐ D 0.003%

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9. What is the formula of Aluminium Carbonate?

- ☐ A $\text{Al}_2(\text{CO}_3)_3$
☐ B Al_2CO_3
☐ C $\text{Al}_2(\text{CO}_3)_2$
☐ D Al_2CO_4

10. Dinitrogen Monoxide decomposes to produce Nitrogen and Oxygen according to the following equation:



Assuming a constant temperature, how many times greater is the volume of the products than the volume of the reactants?

- ☐ A 0.5
☐ B 0.67
☐ C 1.0
☐ D 1.5

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1. Understanding the eBook 4 Self Check Activity Chem Tutor
 - The Rise of Digital Reading 4 Self Check Activity Chem Tutor
 - Advantages of eBooks Over Traditional Books
2. Identifying 4 Self Check Activity Chem Tutor
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 4 Self Check Activity Chem Tutor
 - User-Friendly Interface
4. Exploring eBook Recommendations from 4 Self Check Activity Chem Tutor
 - Personalized Recommendations
 - 4 Self Check Activity Chem Tutor User Reviews and Ratings
 - 4 Self Check Activity Chem Tutor and Bestseller Lists
5. Accessing 4 Self Check Activity Chem Tutor Free and Paid eBooks
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 - Highlighting and Note-Taking 4 Self Check Activity Chem Tutor
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 4 Self Check Activity Chem Tutor
- 9. Balancing eBooks and Physical Books 4 Self Check Activity Chem Tutor
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 4 Self Check Activity Chem Tutor
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 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 4 Self Check Activity Chem Tutor
 - Setting Reading Goals 4 Self Check Activity Chem Tutor
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 4 Self Check Activity Chem Tutor
 - Fact-Checking eBook Content of 4 Self Check Activity Chem Tutor
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
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