

Multiple Choice Questions

for AQA AS Chemistry

FREE WORKSHEETS

Amount of Substance

This test consists of **TEN** multiple choice questions. You should choose **ONE** answer for each question.

1. How many molecules are present in 15.0 g of PH_3 ? (Avogadro number is 6.022×10^{23})

A 2.66×10^{24}
 B 2.65×10^{24}
 C 1.35×10^{24}
 D 1.37×10^{24}

2. What is the mass number of oxygen?

A 8
 B 16
 C 32
 D 64

3. The density of pure water is 1000 g dm^{-3} . What is the concentration of pure water in mol dm^{-3} ?

A 1 mol dm^{-3}
 B 32 mol dm^{-3}
 C 12 mol dm^{-3}
 D 18 mol dm^{-3}

4. Calculate the mass of KOH that should be dissolved in 500 cm^3 water to give a solution with a concentration of 0.6 mol dm^{-3} .

A 0.0673 g
 B 0.87 g
 C 0.16 g
 D 16.8 g

5. A 1 dm^3 sample of Noble gas X contains 0.04471 moles. The density of the gas is 1.784 g dm^{-3} . Which gas does the sample contain?

A Helium
 B Neon
 C Argon
 D Krypton

6. Calculate the concentration of the solution that is formed by adding 50 cm^3 of water to 25 cm^3 of $0.268 \text{ mol dm}^{-3}$ HCl.

A $0.134 \text{ mol dm}^{-3}$
 B $0.0632 \text{ mol dm}^{-3}$
 C 1.24 mol dm^{-3}
 D $3.35 \times 10^{-2} \text{ mol dm}^{-3}$

7. 0.05 moles of an iron salt FeCl_x reacts with AgNO_3 to form 21.5 g of AgCl . What is the value of x ?

A 1
 B 2
 C 3
 D 4

8. A The uncertainty of a balance is $\pm 0.1\text{g}$. What is the % uncertainty if this balance is used to measure 26.7 g of CuSO_4 ?

A 0.4%
 B 0.1%
 C 0.0404%
 D 0.0031%

9. What is the formula of Aluminium Carbonate?

A $\text{Al}_2(\text{CO}_3)_3$
 B AlCO_3
 C Al_2CO_3
 D AlCO_4

10. Dinitrogen Monoxide decomposes to produce Nitrogen and Oxygen according to the following equation:



Assuming a constant temperature, how many times greater is the volume of the products than the volume of the reactants?

A 2
 B 0.67
 C 1.5
 D 10

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