

# Interactive Science Study Guide

## Chapter 8: Our Solar System

### Lesson 3



#### Words To Know

1. **planet** - a large, ball-shaped body that revolves around the sun.
2. **satellite** - the small, roundish bodies, or moons, that revolve around the planets.
3. **solar system** - the sun, the 8 planets, their moons, and other objects orbiting the sun.
4. **inner planets** - the planets closest to the sun (Rocky surfaces)
5. **outer planets** - the planets furthest from the sun (gas surfaces)
6. **asteroid** - a chunk of rock that orbits the sun.
7. **comet** - a frozen object that orbits the sun. (Frozen ice, dust, or rock)

#### Applying Concepts - The Solar System

8. There are 8 planets.
9. Earth rotates on its axis every 24 hours.
10. Earth revolves around the sun, every 365 days.
11. Planets travel in oval-shaped orbits.
12. The sun's gravity holds the planets in their orbits.
13. This is the center of the solar system.
14. Pluto is not considered a planet anymore, it is a dwarf planet.
15. Major greenhouse gases found in space: Methane and Chlorine.
16. Asteroids orbit between Earth and the sun, but sometimes can break off rocks. Comets are made of frozen things.

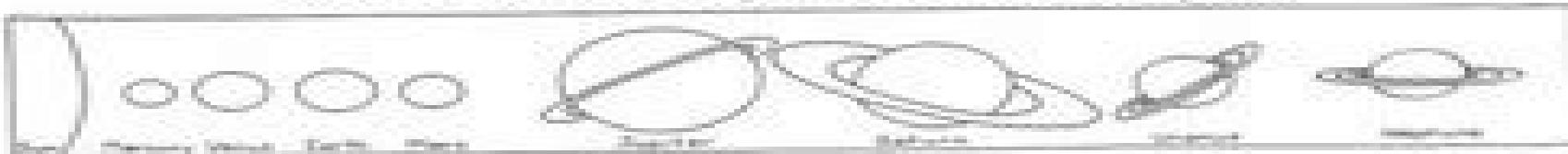


#### The 8 Planets

Know the planets in order from the sun.

Look at these facts about the planets.

Each fact shows what is unique or different about each planet.



17. **Mercury** - the smallest planet.
18. **Venus** - the 2nd planet from the sun.
19. **Earth** - the only planet with water and life.
20. **Mars** - the red planet.
21. **Jupiter** - the largest planet.
22. **Saturn** - the planet with the most rings.
23. **Uranus** - the planet turned on its side.
24. **Neptune** - the planet furthest from the sun. (1 Neptune year = 165 Earth years)

# 3rd Grade Science Study Guide

**Robert R. Redfield**

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co<sub>2</sub> g 3 h 2 o g Δh 283 kJ g 1 c<sub>2</sub>h<sub>6</sub> g

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web thus for reaction 9 2 in which 1 mole of  $h_3o$  and 1 mole of  $oh$  combine to form 2 moles of  $h_2o$   $\Delta h = 55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta h$  is positive if heat is absorbed that is if  $h_{products}$  is greater than  $h_{reactants}$

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