

43. Electropositivity of elements across the periodic table normally

- A. remains constant down the group.
- B. increases across the period.
- C. decreases across the period.
- D. decreases down the group.

44. Consider the following table:

Substance	Melting point / °C	Boiling point / °C
P	-78	-25
Q	-5	40
R	-6	150
S	44	280

Which of the substances is a liquid at room temperature and rapidly evaporates on exposure to air?

- A. P
- B. Q
- C. R
- D. S

45. Which of the following oxides is amphoteric?

- A. Carbon(II) oxide
- B. Nitrogen(IV) oxide
- C. Lead(II) oxide
- D. Calcium oxide

46. Which of the following processes occur during fractional distillation of petroleum?

- A. Condensation and diffusion
- B. Diffusion and evaporation
- C. Diffusion and sublimation
- D. Evaporation and condensation

47. If the molar mass of $X(HCO_3)_2$ is 162 g mol⁻¹, determine the relative atomic mass of X. [H = 1.0, C = 12.0, O = 16.0]

- A. 40
- B. 48
- C. 61
- D. 101

48. Charcoal is used in the decolourization of sugar because of its

- A. absorption property.
- B. amorphous property.
- C. oxidizing property.
- D. adsorption property.

49. The first definition of an element was made by

- A. J. Dalton.
- B. A. Lavoisier.
- C. R. Boyle.
- D. J. J. Thompson.

50. Which of the following scientists formulated the law of conservation of mass?

- A. A. Lavoisier
- B. J. Dalton
- C. R. Boyle
- D. J. Proust

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