

43. Electropositivity of elements across the periodic table normally

A. remains constant down the group.
 B. increases across the period.
 C. decreases across the period.
 D. decreases down the group.

44. Consider the following table:

Substance	Melting point / °C	Boiling point / °C
P	-78	-25
Q	-8	40
R	-6	150
S	44	289

Which of the substances is a liquid at room temperature and rapidly evaporates on exposure to air?

A. P
 B. Q
 C. R
 D. S

45. Which of the following oxides is amphoteric?

A. Carbon(II) oxide
 B. Nitrogen(IV) oxide
 C. Lead(II) oxide
 D. Calcium oxide

46. Which of the following processes occur during fractional distillation of petroleum?

A. Condensation and diffusion
 B. Diffusion and evaporation
 C. Diffusion and sublimation
 D. Evaporation and condensation

47. If the molar mass of $X(\text{HCO}_3)_2$ is 162 g mol^{-1} , determine the relative atomic mass of X. [H = 1.0, C = 12.0, O = 16.0]

A. 40
 B. 48
 C. 61
 D. 101

48. Charcoal is used in the decolourization of sugar because of its

A. absorption property.
 B. amorphous property.
 C. oxidizing property.
 D. adsorption property.

49. The first definition of an element was made by

A. J. Dalton.
 B. A. Lavoisier.
 C. R. Boyle.
 D. J. J. Thompson.

50. Which of the following scientists formulated the law of conservation of mass?

A. A. Lavoisier
 B. J. Dalton
 C. R. Boyle
 D. J. Proust

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Misra, Sanjay, Adewumi, Adewole



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1. Understanding the eBook 2014 Waec Chemistry Paper 3 Answer
 - The Rise of Digital Reading 2014 Waec Chemistry Paper 3 Answer
 - Advantages of eBooks Over Traditional Books
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 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an 2014 Waec Chemistry Paper 3 Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from 2014 Waec Chemistry Paper 3 Answer

- Personalized Recommendations
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- 2014 Waec Chemistry Paper 3 Answer and Bestseller Lists
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 - Highlighting and Note-Taking 2014 Waec Chemistry Paper 3 Answer
 - Interactive Elements 2014 Waec Chemistry Paper 3 Answer
- 8. Staying Engaged with 2014 Waec Chemistry Paper 3 Answer
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2014 Waec Chemistry Paper 3 Answer
- 9. Balancing eBooks and Physical Books 2014 Waec Chemistry Paper 3 Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2014 Waec Chemistry Paper 3 Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 2014 Waec Chemistry Paper 3 Answer
 - Setting Reading Goals 2014 Waec Chemistry Paper 3 Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 2014 Waec Chemistry Paper 3 Answer

- Fact-Checking eBook Content of 2014 Waec Chemistry Paper 3 Answer
- Distinguishing Credible Sources

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- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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