

# Chemistry, Part 1

## Module 14 Notes



### Big Question

How are metals used in chemical processes, reactions, and production of chemical reactions?

### Big Idea

Using chemical analysis, you can identify the metals of a substance or particles of that substance. A metal is often used in manufacturing. It is used in construction as structural analysis with chemical reactions.

### Learning Objectives

After completing this module, students will be able to:

- Use the law of conservation of mass to balance chemical reactions.
- Use chemical reactions to describe processes of materials.
- Compare different metals and reactions.
- Describe the properties of different metals.

### Module 14 Big Idea

#### Learning Objectives

How are metals used in chemical reactions?

Fill in the blanks.

1. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
2. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
3. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.

### Module 14 Big Idea

How are metals used in chemical reactions?

Chemical reactions are used to produce new products from reactants.

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### Module 14 Big Idea

#### Learning Objectives

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