

# Chemistry Review Packet

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## A. Chemistry of the following reactions:



Which [2.0 mol/L] Hydrogen Peroxide reacted with 0.500 g atoms of oxygen. How many grams of water were formed?

A. 1.000 g atoms B. 0.500 g atoms C. 0.250 g atoms D. 0.125 g atoms E. 0.0625 g atoms

12. Which of the following is NOT a sequence of three that form a pH7?

A. H<sub>2</sub>O<sub>2</sub> B. H<sub>2</sub>O<sub>3</sub> C. H<sub>2</sub>O<sub>2</sub><sup>2-</sup> D. H<sub>2</sub>O<sub>2</sub><sup>-</sup> E. H<sub>2</sub>O<sub>2</sub><sup>3-</sup>

13. Which of the following substances would be broken down into simpler substances by physical or chemical means?

A. Water B. Baking soda C. Bleach D. Peroxide E. Oxygen

14. What is the approximate percent mass per cent by mass of Hydrogen in the compound ethyl, C<sub>2</sub>H<sub>5</sub>CO<sub>2</sub> (propanoic acid), to nearest 0.01%? C<sub>2</sub>H<sub>5</sub>CO<sub>2</sub> (propanoic acid) C<sub>2</sub>H<sub>5</sub>CO<sub>2</sub>

A. 40.00% B. 40.00% C. 30.00% D. 20.00% E. 20.00%

15. The empirical formula of a certain hydrocarbon is C<sub>2</sub>H<sub>6</sub>, and it has a molecular weight equal to 60.00. Based on this information, the molecular formula of this hydrocarbon is

A. C<sub>2</sub>H<sub>6</sub> B. C<sub>2</sub>H<sub>12</sub> C. C<sub>2</sub>H<sub>14</sub> D. C<sub>2</sub>H<sub>16</sub> E. C<sub>2</sub>H<sub>18</sub>

16. Which of the following reactions requires an 18.0 g/L Zinc Sulfate, normal solution, percent?

A. 10.00 g C<sub>2</sub>H<sub>5</sub>CO<sub>2</sub> B. 10.00 g H<sub>2</sub>O<sub>2</sub> C. 10.00 g H<sub>2</sub>O<sub>2</sub> D. 10.00 g H<sub>2</sub>O<sub>2</sub>

## B. Chemistry of the reactions:



What is this reaction best classified?

A. Precipitation B. Redox reaction C. Single replacement D. Combination E. Synthesis

17. The following reaction tells you which order standard temperature and pressure?



Which of the following reactions uses oxygen most effectively to remove sulfur dioxide? How many grams of air would be needed?

A. 1.000 g B. 1.000 g C. 1.000 g D. 1.000 g E. 1.000 g

# Answers Chemistry Review Packet

**F Rizvi**

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