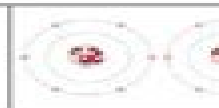


Activity B:
Building
molecules

Get the Gizmo ready:

- Click **Reset**.
- Turn off **Show Lewis diagram**.
- Select **Oxygen**.



Question: How do atoms share more than one pair of electrons?

1. Observe: Like fluorine and most other elements, oxygen atoms are most stable with a full complement of eight valence electrons.
 - A. How many valence electrons does each oxygen atom have now? _____
 - B. How many more electrons does each oxygen atom need to be stable? _____
2. Form a bond: Drag electrons back and forth until the molecule of oxygen (O_2) is stable. Click **Check** to confirm your molecule is stable. Take a snapshot and paste the image into your document (don't forget to label it " O_2 ").

How many *pairs* of shared electrons are there in a stable molecule of oxygen? _____

3. Draw a diagram: Draw a Lewis diagram of the oxygen molecule in the space below at left. To check your work, turn on **Show Lewis diagram**. Draw the correct diagram on the right.

Practice diagram:



Actual:



4. Practice: Create covalent bonds and stable molecules for the remaining substances. Take a snapshot of each completed molecule and add it to your document. Draw Lewis diagrams for each one. (As above, draw the diagram on your own before checking your work.)

Nitrogen



Ammonia



Chlorine



Methane



Water



Formaldehyde



Carbon dioxide



(Activity B continued on next page)

Answer Sheets For Chemistry Gizmos

JG Myers

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Answer Sheets For Chemistry Gizmos

1. Understanding the eBook Answer Sheets For Chemistry Gizmos
 - The Rise of Digital Reading Answer Sheets For Chemistry Gizmos
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Sheets For Chemistry Gizmos
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Sheets For Chemistry Gizmos
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Sheets For Chemistry Gizmos
 - Personalized Recommendations

- Answer Sheets For Chemistry Gizmos User Reviews and Ratings
- Answer Sheets For Chemistry Gizmos and Bestseller Lists
- 5. Accessing Answer Sheets For Chemistry Gizmos Free and Paid eBooks
 - Answer Sheets For Chemistry Gizmos Public Domain eBooks
 - Answer Sheets For Chemistry Gizmos eBook Subscription Services
 - Answer Sheets For Chemistry Gizmos Budget-Friendly Options
- 6. Navigating Answer Sheets For Chemistry Gizmos eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Sheets For Chemistry Gizmos Compatibility with Devices
 - Answer Sheets For Chemistry Gizmos Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Sheets For Chemistry Gizmos
 - Highlighting and Note-Taking Answer Sheets For Chemistry Gizmos
 - Interactive Elements Answer Sheets For Chemistry Gizmos
- 8. Staying Engaged with Answer Sheets For Chemistry Gizmos
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Sheets For Chemistry Gizmos
- 9. Balancing eBooks and Physical Books Answer Sheets For Chemistry Gizmos
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Sheets For Chemistry Gizmos
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Sheets For Chemistry Gizmos
 - Setting Reading Goals Answer Sheets For Chemistry Gizmos
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Sheets For Chemistry Gizmos
 - Fact-Checking eBook Content of Answer Sheets For Chemistry Gizmos

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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