

KEY

Electron Review Questions

Using the equations, solve the following:

$$c = \lambda\nu \quad c = \text{speed of light } 3.0 \times 10^8 \text{ m/s} \quad \nu = \text{frequency} \quad \lambda = \text{wavelength}$$

$$E = h\nu \quad E = \text{energy in Joules} \quad h = \text{Planck constant, } 6.626 \times 10^{-34} \quad \nu = \text{frequency}$$

1. What is the frequency of a wave that has a wavelength of 4×10^{-7} meters? $\nu = \frac{c}{\lambda} = \frac{3.0 \times 10^8}{4 \times 10^{-7}} = 7.50 \times 10^{14} \text{ s}^{-1}$

2. What is the energy of this wave? $E = h\nu = 6.626 \times 10^{-34} \times 7.50 \times 10^{14} = 4.95 \times 10^{-19} \text{ J}$

3. List 5 forms of electromagnetic radiation and what is the lowest energy color of light? Which color of light has the highest energy?

Radio waves, microwaves, visible light, UV, X-rays, Gamma

4. What is the relationship between frequency and wavelength?

It is ~~inversely proportional~~, i.e., as wavelength decreases frequency increases.

5. What is the relationship between energy and frequency?

It is ~~directly proportional~~, as frequency increases, energy increases.

6. Where are the electrons located in the atom?

Well outside the nucleus in the electron cloud

7. Define:

- Ground State - lowest energy state of an atom, ~~atom~~
- Excited State - state with atoms gain energy

8. What is a bright-line or line-emission spectrum of an atom?

A series of specific frequencies and wavelengths. For the light that appears when an element loses its ~~gain~~ through a process.

9. The principle that says no two electrons can have the same set of quantum numbers is Pauli Exclusion Principle.

10. The principle that says electrons will orbit the lowest energy orbitals first is called Aufbau principle.

11. The rule that says electrons will occupy orbitals of equal energy alone, before pairing up is called Hund's Rule.

12. Draw the Aufbau Diagram through 3d.

00000

3p 000

2p 000

3d 000

2d 000

1s 0

13. Name where the blocks located on the periodic table:

a) s block: ~~left column~~

b) p block: ~~right columns~~

c) d block: ~~in the middle blocks; and~~ (B group down)

d) f block: ~~the last at the bottom~~

14. What is the shape of the p orbital? S orbital?

Orbital, spherical

15. Write the electron configuration using orbital notation for sulfur:

$1s \frac{2}{2} 2s \frac{2}{2} 2p \frac{6}{3} 3s \frac{2}{2} 3p \frac{4}{3}$

16. Write the electron configuration for Cadmium using electron configuration notation:

$Cd s^2 2s^2 2p^6 3s^2 3p^6 3d^10 4s^2$

17. Write the electron configuration for lead using noble gas notation:

$Pb [Xe] 6s^2 4f^14 5d^10 6p^2$

18. How many valence electrons does sulfur have?

6, 5

19. Why are copper and chromium exceptions to the Aufbau principle?

They ~~break~~ leave $1s^1$ from $4s$ and ~~leaving~~ fill $3d$ for Cr and ~~anyway~~ fill $3d$ for Cu.

20. Write the electron configuration for the element Sb using the Noble gas abbreviation:

$Sb [Kr] 5s^2 4d^10 5p^3$

Answer Key Unit 3 Review Electrons

Richard Bailey

Answer Key Unit 3 Review Electrons:

The Top Books of the Year Answer Key Unit 3 Review Electrons The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the captivating narratives that have captivated audiences this year. Answer Key Unit 3 Review Electrons : Colleen Hoovers "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and suspenseful novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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