

1.1: Atomic Theory

Write the letter of the correct answer in the space provided.

- 1- O
- 2- O
- 3- C
- 4- A
- 5- O
- 6- O
- 10- A
- 11- A
- 12- B
- 13- A
- 14- B
- 15- C
- 16- O
- 17- A
- 18- O
- 19- A
- 20- C
- 21- C
- 22- A
- 23- B
- 24- A
- 25- B
- 26- B
- 27- B

Use the alphabet below to complete the sentences below.

- 1- Particles
- 2- Electrons
- 3- Atomic Number
- 4- Protons

Match the correct description with the correct term. Write the letter in the space provided.

- 1- a
- 2- a
- 3- c
- 4- b
- 5- b

1. Atom

The smallest unit that can make up an element

2. Electron

Negatively charged particles moving around the nucleus of the atom in a cloud-like region

3. Nucleus

Positively charged. Very dense area at the center of the atom. Contains protons and neutrons

4. electron cloud

A region around the nucleus where electrons are most likely to be found

5. Proton

Positively charged particles in the nucleus

6. Neutron

Particles that have no charge in the nucleus

7. atomic number

Number of the protons in the nucleus

8. isotope

Atoms of the same element with the same number of protons and different number of neutrons

9. mass number

Sum of protons and neutrons in the atom

Atomic Theory Answer Key

Edward C. Gruber



Atomic Theory Answer Key:

CBSE Chemistry Chapterwise Case Study Class 11 Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

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