

Name: \_\_\_\_\_

Period: \_\_\_\_\_

# Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:

number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



Protons:

7

Electrons:

7

Neutrons:

7

Protons:

1

Electrons:

1

Neutrons:

0

Protons:

8

Electrons:

8

Neutrons:

8

## BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
  1. The first energy level can hold 2 electrons
  2. The second energy level can hold 8 electrons
  3. The third energy level can hold 8 electrons

The electrons in the outermost energy level are called:

Valence Electrons

Element: SulfurNumber of Valence Electrons: 6

## LEWIS DIAGRAMS

A lewis diagram shows only the symbol with the Valence electrons



# Atomic Structure Worksheet Answers Key

**Carson-Dellosa Publishing**



## Atomic Structure Worksheet Answers Key:

Teaching and Learning Online Franklin S. Allaire, Jennifer E. Killham, 2023-01-10 Teaching and Learning Online Science for Elementary Grade Levels explores the challenges of teaching science virtually It includes sections on frameworks teacher journeys and lesson plans aligned with Next Generation Science Standards offering tips resources and discussion questions for educators and students

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*Resources in education*, 1989-04

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