

# Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:  
number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



Protons:

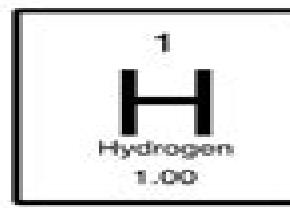
7

Electrons:

7

Neutrons:

7



Protons:

1

Electrons:

1

Neutrons:

0



Protons:

8

Electrons:

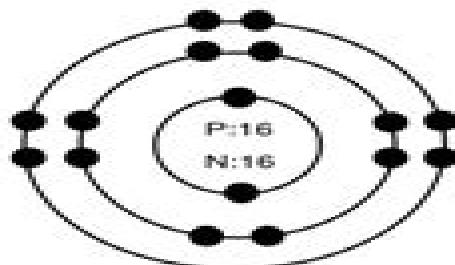
8

Neutrons:

8

## BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
  1. The first energy level can hold 2 electrons
  2. The second energy level can hold 8 electrons
  3. The third energy level can hold 8 electrons

Element: SulfurNumber of Valence Electrons: 6

The electrons in the outermost energy level are called:

Valence Electrons

## LEWIS DIAGRAMS

A Lewis diagram shows only the symbol with the Valence electrons



# Atomic Structure Worksheet Answers Key

**Carson-Dellosa Publishing**

## Atomic Structure Worksheet Answers Key:

Teaching and Learning Online Franklin S. Allaire,Jennifer E. Killham,2023-01-10 Teaching and Learning Online Science for Elementary Grade Levels explores the challenges of teaching science virtually It includes sections on frameworks teacher journeys and lesson plans aligned with Next Generation Science Standards offering tips resources and discussion questions for educators and students

**Chemistry** Carson-Dellosa Publishing,2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

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*Resources in education* ,1989-04 **Oxford Smart Activate Chemistry Teacher Handbook (Ebook)** Alyssa Fox-Charles,2025-06-05 Oxford Smart Activate Chemistry Teacher Handbook Ebook has high aspirations for all budding chemists at KS3 Building on what has been learned at KS2 this handbook helps teachers to plan and deliver lessons that immerse learners in the world of chemistry while developing key knowledge and skills towards GCSE Providing support for all teachers specialists and non subject specialists this handbook contains practical suggestions to reactivate prior knowledge trigger student interest and reflect on learning and progress Links between topics sciences and the wider KS3 curriculum are clearly identified Informed by up to date educational research and tried and tested by Pioneer Schools UK to ensure that the teacher guidance is relevant impactful and promoting current pedagogical practice Oxford Smart Activate is the next evolution of the best selling Activate series from editor and curriculum expert Andrew Chandler Grevatt **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science** ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations

and a wealth of hands on activities take students understanding of science beyond the page and into the world around them  
Now includes even more technology tools and activities to support differentiated instruction

**Chapter Resource 2**

**Chemistry of Life Biology** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004  
Staff,1994      Chemistry Homework Frank Schaffer Publications,Joan DiStasio,1996-03 Includes the periodic table writing  
formulas balancing equations stoichiometry problems and more      Teaching Mineralogy John Brady,David W. Mogk,Dexter  
Perkins,1997      *Holt Chemistry* Ralph Thomas Myers,2004      Science Interactions ,1996      *Prentice Hall Science  
Explorer: Teacher's ed* ,2005      Te HS&T J Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02

*Experimental Chemistry* Robert J. Artz,1982      *Holt Science and Technology 2002* Holt Rinehart & Winston,Holt,  
Rinehart and Winston Staff,2002      Te HS&T a Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02      **Te**

**HS&T 2007 Shrt Crs M** Holt Rinehart & Winston,2007      **Holt Science and Technology** Holt Rinehart &  
Winston,2004-02      **Matter, Building Block of the Universe** ,1993

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